

COMPOUNDS, COMPOSITIONS, AND METHODS

CROSS-REFERENCE TO RELATED PATENT APPLICATIONS

[0001] This application claims the benefit of U.S. Provisional Patent Application number 60/443,911, filed January 30, 2003 and of U.S. Provisional Patent Application number 60/440,873, filed January 17, 2003; each of which is incorporated herein by reference for all purposes.

FIELD OF THE INVENTION

[0002] This invention relates to compounds which are inhibitors of the mitotic kinesin KSP and are useful in the treatment of cellular proliferative diseases, for example cancer, hyperplasias, restenosis, cardiac hypertrophy, immune disorders, fungal disorders, and inflammation.

BACKGROUND OF THE INVENTION

[0003] Among the therapeutic agents used to treat cancer are the taxanes and vinca alkaloids, which act on microtubules. Microtubules are the primary structural element of the mitotic spindle. The mitotic spindle is responsible for distribution of replicate copies of the genome to each of the two daughter cells that result from cell division. It is presumed that disruption of the mitotic spindle by these drugs results in inhibition of cancer cell division, and induction of cancer cell death. However, microtubules form other types of cellular structures, including tracks for intracellular transport in nerve processes. Because these agents do not specifically target mitotic spindles, they have side effects that limit their usefulness.

[0004] Improvements in the specificity of agents used to treat cancer is of considerable interest because of the therapeutic benefits which would be realized if the side effects associated with the administration of these agents could be reduced. Traditionally, dramatic improvements in the treatment of cancer are associated with identification of therapeutic agents acting through novel mechanisms. Examples of this include not only the taxanes, but also the camptothecin class of topoisomerase I inhibitors. From both of these perspectives, mitotic kinesins are attractive targets for new anti-cancer agents.

[0005] Mitotic kinesins are enzymes essential for assembly and function of the mitotic spindle, but are not generally part of other microtubule structures, such as in nerve processes. Mitotic kinesins play essential roles during all phases of mitosis. These enzymes are "molecular motors" that transform energy released by hydrolysis of ATP into mechanical force which drives the directional movement of cellular cargoes along microtubules. The catalytic domain sufficient for this task is a compact structure of approximately 340 amino acids. During mitosis, kinesins organize microtubules into the bipolar structure that is the mitotic spindle. Kinesins mediate movement of chromosomes along spindle microtubules, as well as structural changes in the mitotic spindle associated with specific phases of mitosis. Experimental perturbation of mitotic kinesin function causes malformation or dysfunction of the mitotic spindle, frequently resulting in cell cycle arrest and cell death.

[0006] Among the mitotic kinesins which have been identified is KSP. KSP belongs to an evolutionarily conserved kinesin subfamily of plus end-directed microtubule motors that assemble into bipolar homotetramers consisting of antiparallel homodimers. During mitosis KSP associates with microtubules of the mitotic spindle. Microinjection of antibodies directed against KSP into human cells prevents spindle pole separation during prometaphase, giving rise to monopolar spindles and causing mitotic arrest and induction of programmed cell death. KSP and related kinesins in other, non-human, organisms, bundle antiparallel microtubules and slide them relative to one another, thus forcing the two spindle poles apart. KSP may also mediate in anaphase B spindle elongation and focussing of microtubules at the spindle pole.

[0007] Human KSP (also termed HsEg5) has been described (Blangy, et al., *Cell*, 83:1159-69 (1995); Whitehead, et al., *Arthritis Rheum.*, 39:1635-42 (1996); Galgio et al., *J. Cell Biol.*, 135:339-414 (1996); Blangy, et al., *J Biol. Chem.*, 272:19418-24 (1997); Blangy, et al., *Cell Motil Cytoskeleton*, 40:174-82 (1998); Whitehead and Rattner, *J. Cell Sci.*, 111:2551-61 (1998); Kaiser, et al., *JBC* 274:18925-31 (1999); GenBank accession numbers: X85137, NM004523 and U37426), and a fragment of the KSP gene (TRIP5) has been described (Lee, et al., *Mol Endocrinol.*, 9:243-54 (1995); GenBank accession number L40372). *Xenopus* KSP homologs (Eg5), as well as *Drosophila* KLP61 F/KRP1 30 have been reported.

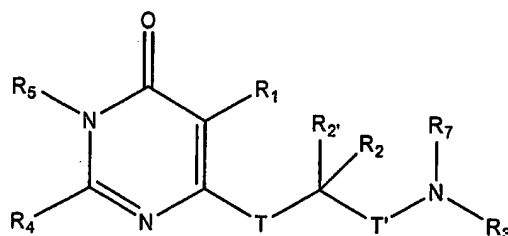
[0008] Mitotic kinesins, including KSP, are attractive targets for the discovery and development of novel antimitotic chemotherapeutics. Accordingly, it is an object of the present invention to provide compounds, compositions and methods useful in the inhibition

of KSP.

SUMMARY OF THE INVENTION

[0009] In accordance with the objects outlined above, the present invention provides compounds that can be used to treat cellular proliferative diseases. The compounds are KSP inhibitors, particularly human KSP inhibitors. The present invention also provides compositions comprising such compounds, and methods utilizing such compounds or compositions, which can be used to treat cellular proliferative diseases.

[0010] In one aspect, the invention relates to methods for treating cellular proliferative diseases, and for treating disorders by inhibiting the activity of KSP. The methods employ one or more compounds represented by Formula I:



Formula I

wherein:

T and T' are independently a covalent bond or optionally substituted lower alkylene;

R₁ is chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, and optionally substituted heteroaralkyl;

R₂ and R₂' are independently chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, and optionally substituted heteroaralkyl; or R₂ and R₂' taken together form an optionally substituted 3- to 7-membered ring;

R₃ is chosen from hydrogen, optionally substituted alkyl-, optionally substituted aryl-, optionally substituted aralkyl-, optionally substituted heteroaryl-, optionally substituted heteroaralkyl-, -C(O)-R₆, and -S(O)₂-R_{6a};

R₄ is independently chosen from hydrogen, optionally substituted alkyl, optionally

substituted alkoxy, hydroxyl, nitro, cyano, dialkylamino, alkylsulfonyl, alkylsulfonamido, alkylthio, carboxyalkyl, carboxamido, aminocarbonyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaralkyl and optionally substituted heteroaryl; and R₅ is hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, or optionally substituted heteroaralkyl; or

R₄ taken together with R₅ form an optionally substituted 5 to 7-membered ring nitrogen-containing heterocycle, which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring;

R₆ is chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, optionally substituted heteroaralkyl, R₁₁O- and R₁₂-NH-;

R_{6a} is chosen from optionally substituted alkyl, optionally substituted aryl, optionally substituted alkylaryl, optionally substituted heteroaryl, optionally substituted alkylheteroaryl, and R₁₂-NH-;

R₇ is chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, and optionally substituted heteroaralkyl;

or R₇ taken together with R₃, and the nitrogen to which they are bound, form an optionally substituted 5- to 12-membered nitrogen-containing heterocycle, which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring;

or R₇ taken together with R₂ form an optionally substituted 5- to 12-membered nitrogen-containing heterocycle, which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring;

R₁₁ is chosen from optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, and optionally substituted heteroaralkyl; and

R₁₂ is chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, and optionally substituted heteroaralkyl;

(Formula I including single stereoisomers and mixtures of stereoisomers);
a pharmaceutically acceptable salt of a compound of Formula I;

a pharmaceutically acceptable solvate of a compound of Formula I; or
a pharmaceutically acceptable solvate of a pharmaceutically acceptable salt of a compound of Formula I.

[0011] In one aspect, the invention relates to methods for treating cellular proliferative diseases and other disorders that can be treated by inhibiting KSP by the administration of a therapeutically effective amount of a compound of Formula I; a pharmaceutically acceptable salt of a compound of Formula I; a pharmaceutically acceptable solvate of a compound of Formula I; or a pharmaceutically acceptable solvate of a pharmaceutically acceptable salt of a compound of Formula I. Such diseases and disorders include cancer, hyperplasia, restenosis, cardiac hypertrophy, immune disorders, fungal disorders and inflammation.

[0012] In another aspect, the invention relates to compounds useful in inhibiting KSP kinesin. The compounds have the structures shown above in Formula I; a pharmaceutically acceptable salt of a compound of Formula I; a pharmaceutically acceptable solvate of a compound of Formula I; or a pharmaceutically acceptable solvate of a pharmaceutically acceptable salt of a compound of Formula I. The invention also relates to pharmaceutical compositions comprising: a therapeutically effective amount of a compound of Formula I; a pharmaceutically acceptable salt of a compound of Formula I; a pharmaceutically acceptable solvate of a compound of Formula I; or a pharmaceutically acceptable solvate of a pharmaceutically acceptable salt of a compound of Formula I; and one or more pharmaceutical excipients. In another aspect, the composition further comprises a chemotherapeutic agent other than a compound of the present invention.

[0013] In an additional aspect, the present invention provides methods of screening for compounds that will bind to a KSP kinesin, for example compounds that will displace or compete with the binding of a compound of the invention. The methods comprise combining a labeled compound of the invention, a KSP kinesin, and at least one candidate agent and determining the binding of the candidate agent to the KSP kinesin.

[0014] In a further aspect, the invention provides methods of screening for modulators of KSP kinesin activity. The methods comprise combining a compound of the invention, a KSP kinesin, and at least one candidate agent and determining the effect of the candidate agent on the KSP kinesin activity.

DETAILED DESCRIPTION OF THE INVENTION**Definitions**

[0015] As used in the present specification, the following words and phrases are generally intended to have the meanings as set forth below, except to the extent that the context in which they are used indicates otherwise. The following abbreviations and terms have the indicated meanings throughout:

Ac	=	acetyl
BNB	=	4-bromomethyl-3-nitrobenzoic acid
Boc	=	t-butyloxy carbonyl
Bu	=	butyl
c-	=	cyclo
CBZ	=	carbobenzoxo = benzyloxycarbonyl
DBU	=	diazabicyclo[5.4.0]undec-7-ene
DCM	=	dichloromethane = methylene chloride = CH_2Cl_2
DCE	=	dichloroethane
DEAD	=	diethyl azodicarboxylate
DIC	=	diisopropylcarbodiimide
DIEA	=	N,N-diisopropylethylamine
DMAP	=	4-N,N-dimethylaminopyridine
DMF	=	N,N-dimethylformamide
DMSO	=	dimethyl sulfoxide
DVB	=	1,4-divinylbenzene
EEDQ	=	2-ethoxy-1-ethoxycarbonyl-1,2-dihydroquinoline
Et	=	ethyl
Fmoc	=	9-fluorenylmethoxycarbonyl
GC	=	gas chromatography
HATU	=	O-(7-Azabenzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate
HMDS	=	hexamethyldisilazane
HOAc	=	acetic acid
HOBt	=	hydroxybenzotriazole
Me	=	methyl
mesyl	=	methanesulfonyl

MTBE	=	methyl t-butyl ether
NMO	=	N-methylmorpholine oxide
PEG	=	polyethylene glycol
Ph	=	phenyl
PhOH	=	phenol
PfP	=	pentafluorophenol
PPTS	=	pyridinium p-toluenesulfonate
Py	=	pyridine
PyBroP	=	bromo-tris-pyrrolidino-phosphonium hexafluorophosphate
rt	=	room temperature
sat'd	=	saturated
s-	=	secondary
t-	=	tertiary
TBDMS	=	t-butyldimethylsilyl
TES	=	triethylsilyl
TFA	=	trifluoroacetic acid
THF	=	tetrahydrofuran
TMOF	=	trimethyl orthoformate
TMS	=	trimethylsilyl
tosyl	=	p-toluenesulfonyl
Trt	=	triphenylmethyl

[0016] **Alkyl** is intended to include linear, branched, or cyclic aliphatic hydrocarbon structures and combinations thereof, which structures may be saturated or unsaturated.

Lower-alkyl refers to alkyl groups of from 1 to 5 carbon atoms, preferably from 1 to 4 carbon atoms. Examples of lower-alkyl groups include methyl-, ethyl-, propyl-, isopropyl-, butyl-, s- and t-butyl and the like. Preferred alkyl groups are those of C₂₀ or below. More preferred alkyl groups are those of C₁₃ or below. **Cycloalkyl** is a subset of alkyl and includes cyclic aliphatic hydrocarbon groups of from 3 to 13 carbon atoms. Examples of cycloalkyl groups include c-propyl-, c-butyl-, c-pentyl-, norbornyl-, adamantyl and the like. **Cycloalkyl-alkyl-** is another subset of alkyl and refers to cycloalkyl attached to the parent structure through a non-cyclic alkyl-. Examples of cycloalkyl-alkyl- include cyclohexylmethyl-,

cyclopropylmethyl-, cyclohexylpropyl-, and the like. In this application, alkyl includes alkanyl-, alkenyl and alkynyl residues; it is intended to include vinyl-, allyl-, isoprenyl and the like. When an alkyl residue having a specific number of carbons is named, all geometric isomers having that number of carbons are intended to be encompassed; thus, for example, "butyl" is meant to include n-butyl-, sec-butyl-, isobutyl and t-butyl-; "propyl" includes n-propyl-, isopropyl-, and c-propyl-.

[0017] **Alkylene-, alkenylene-, and alkynylene-** are other subsets of alkyl-, including the same residues as alkyl-, but having two points of attachment within a chemical structure. Examples of alkylene include ethylene ($-\text{CH}_2\text{CH}_2-$), propylene ($-\text{CH}_2\text{CH}_2\text{CH}_2-$), dimethylpropylene ($-\text{CH}_2\text{C}(\text{CH}_3)_2\text{CH}_2-$) and cyclohexylpropylene ($-\text{CH}_2\text{CH}_2\text{CH}(\text{C}_6\text{H}_{13})-$). Likewise, examples of alkenylene include ethenylene ($-\text{CH}=\text{CH}-$), propenylene ($-\text{CH}=\text{CH}-\text{CH}_2-$), and cyclohexylpropenylene ($-\text{CH}=\text{CHCH}(\text{C}_6\text{H}_{13})-$). Examples of alkynylene include ethynylene ($-\text{C}\equiv\text{C}-$) and propynylene ($-\text{CH}\equiv\text{CH}-\text{CH}_2-$).

[0018] **Cycloalkenyl** is a subset of alkyl and includes unsaturated cyclic hydrocarbon groups of from 3 to 13 carbon atoms. Examples of cycloalkenyl groups include c-hexenyl-, c-pentenyl and the like.

[0019] **Alkoxy or alkoxyl** refers to an alkyl group, preferably including from 1 to 8 carbon atoms, of a straight, branched, or cyclic configuration, or a combination thereof, attached to the parent structure through an oxygen (i.e., the group alkyl-O-). Examples include methoxy-, ethoxy-, propoxy-, isopropoxy-, cyclopropyloxy-, cyclohexyloxy- and the like. **Lower-alkoxy** refers to alkoxy groups containing one to four carbons.

[0020] **Acyl** refers to groups of from 1 to 8 carbon atoms of a straight, branched, or cyclic configuration or a combination thereof, attached to the parent structure through a carbonyl functionality. Such groups may be saturated or unsaturated, and aliphatic or aromatic. One or more carbons in the acyl residue may be replaced by oxygen, nitrogen (e.g., carboxamido), or sulfur as long as the point of attachment to the parent remains at the carbonyl. Examples include acetyl-, benzoyl-, propionyl-, isobutyryl-, oxalyl-, t-butoxycarbonyl-, benzyloxycarbonyl-, morpholinylcarbonyl-, and the like. **Lower-acyl** refers to acyl groups containing one to four carbons.

[0021] **Amino** refers to the group $-\text{NH}_2$. The term "substituted amino" refers to the group $-\text{NHR}$ or $-\text{NRR}$ where each R is independently selected from the group: optionally substituted alkyl-, optionally substituted alkoxy, optionally substituted aminocarbonyl-, optionally substituted aryl-, optionally substituted heteroaryl-, optionally substituted

heterocyclyl-, acyl-, alkoxy carbonyl-, sulfanyl-, sulfinyl and sulfonyl-, e.g., diethylamino, methylsulfonylamino, furanyl-oxy-sulfonamino. Substituted amino includes the groups – NR^cCOR^b , $-\text{NR}^c\text{CO}_2\text{R}^a$, and $-\text{NR}^c\text{CONR}^b\text{R}^c$, where

R^a is an optionally substituted C_1 - C_6 alkyl-, aryl-, heteroaryl-, aryl- C_1 - C_4 alkyl-, or heteroaryl- C_1 - C_4 alkyl- group;

R^b is H or optionally substituted C_1 - C_6 alkyl-, aryl-, heteroaryl-, aryl- C_1 - C_4 alkyl-, or heteroaryl- C_1 - C_4 alkyl- group; and

R^c is hydrogen or C_1 - C_4 alkyl-; and

where each optionally substituted R^b group is independently unsubstituted or substituted with one or more substituents independently selected from C_1 - C_4 alkyl-, aryl-, heteroaryl-, aryl- C_1 - C_4 alkyl-, heteroaryl- C_1 - C_4 alkyl-, C_1 - C_4 haloalkyl-, $-\text{OC}_1$ - C_4 alkyl-, $-\text{OC}_1$ - C_4 alkylphenyl, $-\text{C}_1$ - C_4 alkyl-OH, $-\text{OC}_1$ - C_4 haloalkyl, halogen, -OH, - NH_2 , $-\text{C}_1$ - C_4 alkyl- NH_2 , $-\text{N}(\text{C}_1$ - C_4 alkyl)(C_1 - C_4 alkyl), $-\text{NH}(\text{C}_1$ - C_4 alkyl), $-\text{N}(\text{C}_1$ - C_4 alkyl)(C_1 - C_4 alkylphenyl), $-\text{NH}(\text{C}_1$ - C_4 alkylphenyl), cyano, nitro, oxo (as a substituent for heteroaryl), $-\text{CO}_2\text{H}$, $-\text{C}(\text{O})\text{OC}_1$ - C_4 alkyl, $-\text{CON}(\text{C}_1$ - C_4 alkyl)(C_1 - C_4 alkyl), $-\text{CONH}(\text{C}_1$ - C_4 alkyl), $-\text{CONH}_2$, $-\text{NHC}(\text{O})(\text{C}_1$ - C_4 alkyl), $-\text{NHC}(\text{O})(\text{phenyl})$, $-\text{N}(\text{C}_1$ - C_4 alkyl) $\text{C}(\text{O})(\text{C}_1$ - C_4 alkyl), $-\text{N}(\text{C}_1$ - C_4 alkyl) $\text{C}(\text{O})(\text{phenyl})$, $-\text{C}(\text{O})\text{C}_1$ - C_4 alkyl, $-\text{C}(\text{O})\text{C}_1$ - C_4 phenyl, $-\text{C}(\text{O})\text{C}_1$ - C_4 haloalkyl, $-\text{OC}(\text{O})\text{C}_1$ - C_4 alkyl, $-\text{SO}_2(\text{C}_1$ - C_4 alkyl), $-\text{SO}_2(\text{phenyl})$, $-\text{SO}_2(\text{C}_1$ - C_4 haloalkyl), $-\text{SO}_2\text{NH}_2$, $-\text{SO}_2\text{NH}(\text{C}_1$ - C_4 alkyl), $-\text{SO}_2\text{NH}(\text{phenyl})$, $-\text{NHSO}_2(\text{C}_1$ - C_4 alkyl), $-\text{NHSO}_2(\text{phenyl})$, and $-\text{NHSO}_2(\text{C}_1$ - C_4 haloalkyl).

[0022] **Antimitotic** refers to a drug for inhibiting or preventing mitosis, for example, by causing metaphase arrest. Some antitumour drugs block proliferation and are considered antimitotics.

[0023] **Aryl and heteroaryl** mean a 5- or 6-membered aromatic or heteroaromatic ring containing 0 or 1-4 heteroatoms, respectively, selected from O, N, or S; a bicyclic 9- or 10-membered aromatic or heteroaromatic ring system containing 0 or 1-4 (or more) heteroatoms, respectively, selected from O, N, or S; or a tricyclic 12- to 14-membered aromatic or heteroaromatic ring system containing 0 or 1-4 (or more) heteroatoms, respectively, selected from O, N, or S. The aromatic 6- to 14-membered carbocyclic rings include, e.g., phenyl-, naphthyl-, indanyl-, tetralinyl-, and fluorenyl and the 5- to 10-membered aromatic heterocyclic rings include, e.g., imidazolyl-, pyridinyl-, indolyl-, thienyl-, benzopyranonyl-, thiazolyl-, furanyl-, benzimidazolyl-, quinolinyl-, isoquinolinyl-, quinoxalinyl-, pyrimidinyl-, pyrazinyl-, tetrazolyl and pyrazolyl-.

[0024] **Aralkyl-** refers to a residue in which an aryl moiety is attached to the parent structure via an alkyl residue. Examples include benzyl-, phenethyl-, phenylvinyl-, phenylallyl and the like. **Heteroaralkyl-** refers to a residue in which a heteroaryl moiety is attached to the parent structure via an alkyl residue. Examples include furanylmethyl-, pyridinylmethyl-, pyrimidinylethyl and the like.

[0025] **Aralkoxy-** refers to the group -O-aralkyl. Similarly, **heteroaralkoxy-** refers to the group -O-heteroaralkyl-; **aryloxy-** refers to the group -O-aryl-; **acyloxy-** refers to the group -O-acyl-; **heteroaryloxy-** refers to the group -O-heteroaryl-; and **heterocycloxy-** refers to the group -O-heterocyclyl (i.e., aralkyl-, heteroaralkyl-, aryl-, acyl-, heterocyclyl-, or heteroaryl is attached to the parent structure through an oxygen).

[0026] **Carboxyalkyl-** refers to the group -alkyl-COOH.

[0027] **Aminocarbonyl** refers to the group -CONR^bR^c, where

R^b is H or optionally substituted C₁-C₆ alkyl-, aryl-, heteroaryl-, aryl-C₁-C₄ alkyl-, or heteroaryl-C₁-C₄ alkyl- group; and

R^c is hydrogen or C₁-C₄ alkyl-; and

where each optionally substituted R^b group is independently unsubstituted or substituted with one or more substituents independently selected from C₁-C₄ alkyl-, aryl-, heteroaryl-, aryl-C₁-C₄ alkyl-, heteroaryl-C₁-C₄ alkyl-, C₁-C₄ haloalkyl-, -OC₁-C₄ alkyl-, -OC₁-C₄ alkylphenyl, -C₁-C₄ alkyl-OH, -OC₁-C₄ haloalkyl, halogen, -OH, -NH₂, -C₁-C₄ alkyl-NH₂, -N(C₁-C₄ alkyl)(C₁-C₄ alkyl), -NH(C₁-C₄ alkyl), -N(C₁-C₄ alkyl)(C₁-C₄ alkylphenyl), -NH(C₁-C₄ alkylphenyl), cyano, nitro, oxo (as a substituent for heteroaryl), -CO₂H, -C(O)OC₁-C₄ alkyl, -CON(C₁-C₄ alkyl)(C₁-C₄ alkyl), -CONH(C₁-C₄ alkyl), -CONH₂, -NHC(O)(C₁-C₄ alkyl), -NHC(O)(phenyl), -N(C₁-C₄ alkyl)C(O)(C₁-C₄ alkyl), -N(C₁-C₄ alkyl)C(O)(phenyl), -C(O)C₁-C₄ alkyl, -C(O)C₁-C₄ phenyl, -C(O)C₁-C₄ haloalkyl, -OC(O)C₁-C₄ alkyl, -SO₂(C₁-C₄ alkyl), -SO₂(phenyl), -SO₂(C₁-C₄ haloalkyl), -SO₂NH₂, -SO₂NH(C₁-C₄ alkyl), -SO₂NH(phenyl), -NHSO₂(C₁-C₄ alkyl), -NHSO₂(phenyl), and -NHSO₂(C₁-C₄ haloalkyl). Aminocarbonyl is meant to include carbamoyl-, lower-alkyl carbamoyl-, benzylcarbamoyl-, phenylcarbamoyl-, methoxymethyl-carbamoyl-, and the like.

[0028] **Halogen** or **halo** refers to fluorine, chlorine, bromine or iodine. Fluorine, chlorine and bromine are preferred. Dihaloaryl-, dihaloalkyl-, trihaloaryl etc. refer to aryl and alkyl substituted with the designated plurality of halogens (here, 2, 2 and 3, respectively), but not necessarily a plurality of the same halogen; thus 4-chloro-3-fluorophenyl is within the

scope of dihaloaryl-.

[0029] **Heterocyclyl** means a cycloalkyl or aryl residue in which one to four of the carbons is replaced by a heteroatom such as oxygen, nitrogen or sulfur. Examples of heterocycles that fall within the scope of the invention include azetidiny-, imidazolinyl-, pyrrolidinyl-, pyrazolyl-, pyrrolyl-, indolyl-, quinolinyl-, isoquinolinyl-, tetrahydroisoquinolinyl-, benzofuranyl-, benzodioxanyl-, benzodioxyl (commonly referred to as methylenedioxyphenyl-, when occurring as a substituent), tetrazolyl-, morpholinyl-, thiazolyl-, pyridinyl-, pyridazinyl-, piperidinyl-, pyrimidinyl-, thienyl-, furanyl-, oxazolyl-, oxazolinyl-, isoxazolyl-, dioxanyl-, tetrahydrofuranyl and the like. "N-heterocyclyl" refers to a nitrogen-containing heterocycle. The term heterocyclyl encompasses heteroaryl-, which is a subset of heterocyclyl-. Examples of N-heterocyclyl residues include azetidiny-, 4-morpholinyl-, 4-thiomorpholinyl-, 1-piperidinyl-, 1-pyrrolidinyl-, 3-thiazolidinyl-, piperazinyl and 4-(3,4-dihydrobenzoxazinyl). Examples of substituted heterocyclyl include 4-methyl-1-piperazinyl and 4-benzyl-1-piperidinyl-.

[0030] A **leaving group or atom** is any group or atom that will, under the reaction conditions, cleave from the starting material, thus promoting reaction at a specified site. Suitable examples of such groups unless otherwise specified are halogen atoms, mesyloxy, p-nitrobenzensulphonyloxy and tosyloxy groups.

[0031] **Optional or optionally** means that the subsequently described event or circumstance may or may not occur, and that the description includes instances where said event or circumstances occurs and instances in which it does not. For example, "optionally substituted alkyl" includes "alkyl" and "substituted alkyl" as defined herein. It will be understood by those skilled in the art with respect to any group containing one or more substituents that such groups are not intended to introduce any substitution or substitution patterns that are sterically impractical and/or synthetically non-feasible and/or inherently unstable.

[0032] **Substituted alkoxy** refers to alkoxy wherein the alkyl constituent is substituted (i.e., -O-(substituted alkyl)). One suitable substituted alkoxy group is "polyalkoxy" or -O-(optionally substituted alkylene)-(optionally substituted alkoxy), and includes groups such as -OCH₂CH₂OCH₃, and residues of glycol ethers such as polyethyleneglycol, and -O(CH₂CH₂O)_xCH₃, where x is an integer of about 2-20, preferably about 2-10, and more preferably about 2-5. Another suitable substituted alkoxy group is hydroxyalkoxy or -OCH₂(CH₂)_yOH, where y is an integer of about 1-10, preferably about 1-4.

[0033] **Substituted-** alkyl-, aryl-, and heteroaryl- refer respectively to alkyl-, aryl-, and heteroaryl wherein one or more (up to about 5, preferably up to about 3) hydrogen atoms are replaced by a substituent independently selected from the group: $-R^a$, $-OR^b$, $-O(C_1-C_2 \text{ alkyl})O-$ (e.g., ethylenedioxy or methylenedioxy), $-SR^b$, guanidine, guanidine wherein one or more of the guanidine hydrogens are replaced with a lower-alkyl group, $-NR^bR^c$, halogen, cyano, nitro, $-COR^b$, $-CO_2R^b$, $-CONR^bR^c$, $-OCOR^b$, $-OCO_2R^a$, $-OCONR^bR^c$, $-NR^cCOR^b$, $-NR^cCO_2R^a$, $-NR^cCONR^bR^c$, $-CO_2R^b$, $-CONR^bR^c$, $-NR^cCOR^b$, $-SOR^a$, $-SO_2R^a$, $-SO_2NR^bR^c$, and $-NR^cSO_2R^a$,

where R^a is an optionally substituted C_1-C_6 alkyl-, aryl-, heteroaryl-, aryl- C_1-C_4 alkyl-, or heteroaryl- C_1-C_4 alkyl- group,

R^b is H or optionally substituted C_1-C_6 alkyl-, aryl-, heteroaryl-, aryl- C_1-C_4 alkyl-, or heteroaryl- C_1-C_4 alkyl- group;

R^c is hydrogen or C_1-C_4 alkyl-;

where each optionally substituted R^a group and R^b group is independently unsubstituted or substituted with one or more substituents independently selected from C_1-C_4 alkyl-, aryl-, heteroaryl-, aryl- C_1-C_4 alkyl-, heteroaryl- C_1-C_4 alkyl-, C_1-C_4 haloalkyl-, $-OC_1-C_4$ alkyl-, $-OC_1-C_4$ alkylphenyl-, $-C_1-C_4$ alkyl-OH, $-OC_1-C_4$ haloalkyl-, halogen, $-OH$, $-NH_2$, $-C_1-C_4$ alkyl- NH_2 , $-N(C_1-C_4 \text{ alkyl})(C_1-C_4 \text{ alkyl})$, $-NH(C_1-C_4 \text{ alkyl})$, $-N(C_1-C_4 \text{ alkyl})(C_1-C_4 \text{ alkylphenyl})$, $-NH(C_1-C_4 \text{ alkylphenyl})$, cyano, nitro, oxo (as a substituent for heteroaryl), $-CO_2H$, $-C(O)OC_1-C_4 \text{ alkyl-}$, $-CON(C_1-C_4 \text{ alkyl})(C_1-C_4 \text{ alkyl})$, $-CONH(C_1-C_4 \text{ alkyl})$, $-CONH_2$, $-NHC(O)(C_1-C_4 \text{ alkyl})$, $-NHC(O)(\text{phenyl})$, $-N(C_1-C_4 \text{ alkyl})C(O)(C_1-C_4 \text{ alkyl})$, $-N(C_1-C_4 \text{ alkyl})C(O)(\text{phenyl})$, $-C(O)C_1-C_4 \text{ alkyl-}$, $-C(O)C_1-C_4 \text{ phenyl-}$, $-C(O)C_1-C_4 \text{ haloalkyl-}$, $-OC(O)C_1-C_4 \text{ alkyl-}$, $-SO_2(C_1-C_4 \text{ alkyl})$, $-SO_2(\text{phenyl})$, $-SO_2(C_1-C_4 \text{ haloalkyl})$, $-SO_2NH_2$, $-SO_2NH(C_1-C_4 \text{ alkyl})$, $-SO_2NH(\text{phenyl})$, $-NHSO_2(C_1-C_4 \text{ alkyl})$, $-NHSO_2(\text{phenyl})$, and $-NHSO_2(C_1-C_4 \text{ haloalkyl})$. In the compounds of Formula I where T and/or T' are substituted alkylene, the term "substituted" also refers to alkylene groups where one or more (up to about 3, particularly 1) carbon atoms are replaced by a heteroatom independently selected from O, N or S, such as $-CH_2-S-CH_2-$.

[0034] **Sulfanyl** refers to the groups: $-S$ -(optionally substituted alkyl), $-S$ -(optionally substituted aryl), $-S$ -(optionally substituted heteroaryl), and $-S$ -(optionally substituted heterocyclyl).

[0035] **Sulfinyl** refers to the groups: $-S(O)-H$, $-S(O)$ -(optionally substituted alkyl),

-S(O)-(optionally substituted aryl), -S(O)-(optionally substituted heteroaryl), -S(O)-(optionally substituted heterocyclyl); and -S(O)-(optionally substituted amino).

[0036] **Sulfonyl** refers to the groups: -S(O₂)-H, -S(O₂)-(optionally substituted alkyl), -S(O₂)-(optionally substituted aryl), -S(O₂)-(optionally substituted heteroaryl), -S(O₂)-(optionally substituted heterocyclyl), -S(O₂)-(optionally substituted alkoxy), -S(O₂)-(optionally substituted aryloxy), -S(O₂)-(optionally substituted heteroaryloxy), -S(O₂)-(optionally substituted heterocyclyloxy); and -S(O₂)-(optionally substituted amino).

[0037] **Pharmaceutically acceptable salts** refers to those salts that retain the biological effectiveness of the free compound and that are not biologically undesirable or unsuitable for pharmaceutical use, formed with a suitable acid or base, and includes pharmaceutically acceptable acid addition salts and base addition salts. **Pharmaceutically acceptable acid addition salts** include those derived from inorganic acids such as hydrochloric acid, hydrobromic acid, sulfuric acid, nitric acid, phosphoric acid and the like, and those derived from organic acids such as acetic acid, propionic acid, glycolic acid, pyruvic acid, oxalic acid, maleic acid, malonic acid, succinic acid, fumaric acid, tartaric acid, citric acid, benzoic acid, cinnamic acid, mandelic acid, methanesulfonic acid, ethanesulfonic acid, p-toluenesulfonic acid, salicylic acid and the like.

[0038] **Pharmaceutically acceptable base addition salts** include those derived from inorganic bases such as sodium, potassium, lithium, ammonium, calcium, magnesium, iron, zinc, copper, manganese, aluminum salts and the like. Particular embodiments are the ammonium, potassium, sodium, calcium, and magnesium salts. Base addition salts also include those derived from pharmaceutically acceptable organic non-toxic bases, including salts of primary, secondary, and tertiary amines, substituted amines including naturally occurring substituted amines, cyclic amines and basic ion exchange resins, such as isopropylamine, trimethylamine, diethylamine, triethylamine, tripropylamine, and ethanolamine.

[0039] **Protecting group** has the meaning conventionally associated with it in organic synthesis, i.e. a group that selectively blocks one or more reactive sites in a multifunctional compound such that a chemical reaction can be carried out selectively on another unprotected reactive site and such that the group can readily be removed after the selective reaction is complete. A variety of protecting groups are disclosed, for example, in T.H. Greene and P. G. M. Wuts, *Protective Groups in Organic Synthesis*, Third Edition, John Wiley & Sons, New York (1999), which is incorporated herein by reference in its entirety. For example, a

hydroxy protected form is where at least one of the hydroxyl groups present in a compound is protected with a hydroxy protecting group. Likewise, amines and other reactive groups may similarly be protected.

[0040] **Solvate** refers to the compound formed by the interaction of a solvent and a compound of Formula I or salt thereof. Suitable solvates of the compounds of the Formula I or a salt thereof are pharmaceutically acceptable solvates including hydrates.

[0041] Many of the compounds described herein contain one or more asymmetric centers (e.g. the carbon to which R_2 and $R_{2'}$ are attached where R_2 differs from $R_{2'}$) and may thus give rise to enantiomers, diastereomers, and other stereoisomeric forms that may be defined, in terms of absolute stereochemistry, as (R)- or (S)-. The present invention is meant to include all such possible isomers, including racemic mixtures, optically pure forms and intermediate mixtures. Optically active (R)- and (S)- isomers may be prepared using chiral synthons or chiral reagents, or resolved using conventional techniques. When the compounds described herein contain olefinic double bonds or other centers of geometric asymmetry, and unless specified otherwise, it is intended that the compounds include both E and Z geometric isomers. Likewise, all tautomeric forms and rotational isomers are also intended to be included.

[0042] When desired, the R- and S-isomers may be resolved by methods known to those skilled in the art, for example by formation of diastereoisomeric salts or complexes which may be separated, for example, by crystallization; via formation of diastereoisomeric derivatives which may be separated, for example, by crystallization, gas-liquid or liquid chromatography; selective reaction of one enantiomer with an enantiomer-specific reagent, for example enzymatic oxidation or reduction, followed by separation of the modified and unmodified enantiomers; or gas-liquid or liquid chromatography in a chiral environment, for example on a chiral support, such as silica with a bound chiral ligand or in the presence of a chiral solvent. It will be appreciated that where the desired enantiomer is converted into another chemical entity by one of the separation procedures described above, a further step may be required to liberate the desired enantiomeric form. Alternatively, specific enantiomer may be synthesized by asymmetric synthesis using optically active reagents, substrates, catalysts or solvents, or by converting one enantiomer to the other by asymmetric transformation.

Compounds of the Present Invention

[0043] The present invention is directed to a class of novel compounds that are inhibitors of one or more mitotic kinesins. By inhibiting mitotic kinesins, but not other kinesins (e.g., transport kinesins), specific inhibition of cellular proliferation is accomplished.

While not intending to be bound by any theory, the present invention capitalizes on the finding that perturbation of mitotic kinesin function causes malformation or dysfunction of mitotic spindles, frequently resulting in cell cycle arrest and cell death. According to one embodiment of the invention, the compounds described herein inhibit the mitotic kinesin, KSP, particularly human KSP. In another embodiment, the compounds inhibit the mitotic kinesin, KSP, as well as modulating one or more of the human mitotic kinesins selected from the group consisting of HSET (see, U.S. Patent No. 6,361,993, which is incorporated herein by reference); MCAK (see, U.S. Patent No. 6,331,424, which is incorporated herein by reference); CENP-E (see, PCT Publication No. WO 99/13061, which is incorporated herein by reference); Kif4 (see, U.S. Patent No. 6,440,684, which is incorporated herein by reference); MKLP1 (see, U.S. Patent No. 6,448,025, which is incorporated herein by reference); Kif15 (see, U.S. Patent No. 6,355,466, which is incorporated herein by reference); Kid (see, U.S. Patent No. 6,387,644, which is incorporated herein by reference); Mpp1, CMKrp, KinI-3 (see, U.S. Patent No. 6,461,855, which is incorporated herein by reference); Kip3a (see, PCT Publication No. WO 01/96593, which is incorporated herein by reference); Kip3d (see, U.S. Patent No. 6,492,151, which is incorporated herein by reference); and RabK6.

[0044] The methods of inhibiting a mitotic kinesin comprise contacting an inhibitor of the invention with a kinesin, particularly a human kinesin, more particularly, human KSP or fragments and variants thereof. The inhibition can be of the ATP hydrolysis activity of the KSP kinesin and/or the mitotic spindle formation activity, such that the mitotic spindles are disrupted. Meiotic spindles may also be disrupted.

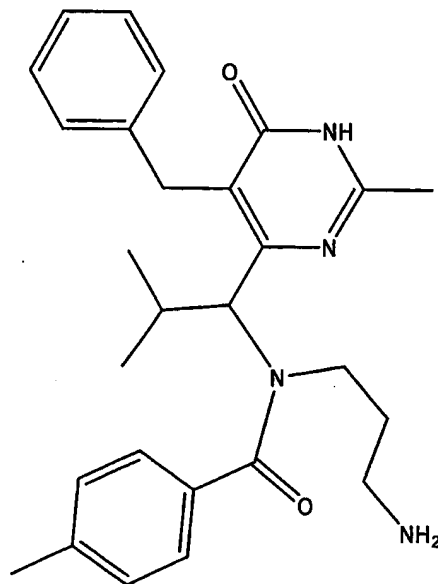
[0045] The present invention provides inhibitors of mitotic kinesins, in particular KSP and especially human KSP, for the treatment of disorders associated with cell proliferation. The compounds, compositions and methods described herein can differ in their selectivity and are used to treat diseases of cellular proliferation, including, but not limited to cancer, hyperplasias, restenosis, cardiac hypertrophy, immune disorders, fungal disorders and inflammation.

[0046] Accordingly, the present invention relates to methods employing compounds represented by Formula I:

Nomenclature

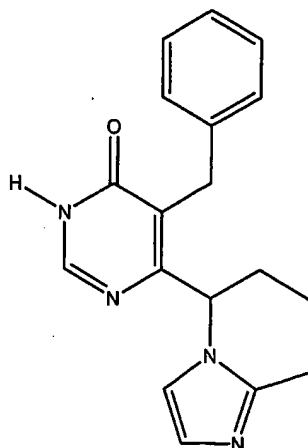
[0047] The compounds of Formula I are named and numbered as described below.

For example, the compound:



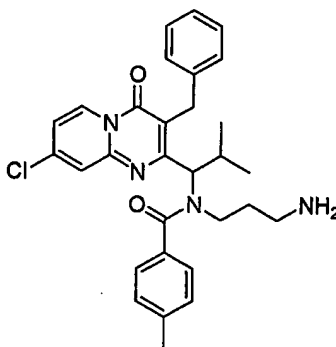
i.e., the compound according to Formula I where T and T' are absent; R₁ is benzyl, R₂ is propyl (or i-propyl), R₂' is hydrogen; R₃ is -COR₆; R₇ is aminopropyl; R₅ is hydrogen; R₄ is methyl; and R₆ is p-tolyl can be named N-(3-amino-propyl)-N-[1-(5-benzyl-1-methyl-6-oxo-1,6-dihydro-pyrimidin-4-yl)-2-methyl-propyl]-4-methyl-benzamide.

[0048] Likewise, the compound



i.e., the compound according to Formula I where T and T' are absent; R₁ is benzyl, R₂ is ethyl, R_{2'} is hydrogen; R₃ taken together with R₇ is a substituted imidazolyl; and R₄ and R₅ are hydrogen can be named 5-benzyl-6-[1-(2-methyl-imidazol-1-yl)-propyl]-3H-pyrimidin-4-one.

[0049] Likewise, the compound having the structure



i.e., the compound according to Formula I where T and T' are absent; R₁ is benzyl; R₂ is propyl (or i-propyl); R_{2'} is hydrogen; R₃ is -COR₆; R₆ is p-tolyl; R₇ is 3-aminopropyl; and R₄ and R₅ taken together form a substituted pyridinyl ring can be named N-(3-amino-propyl)-N-[1-(3-benzyl-8-chloro-4-oxo-4H-pyrido[1,2-a]pyrimidin-2-yl)-2-methyl-propyl]-4-methylbenzamide.

Synthetic Reaction Parameters

[0050] The compounds of Formula I can be prepared by following the procedures described with reference to the Reaction Schemes below.

[0051] Unless specified otherwise, the terms "solvent", "inert organic solvent" or "inert solvent" mean a solvent inert under the conditions of the reaction being described in conjunction therewith [including, for example, benzene, toluene, acetonitrile, tetrahydrofuran ("THF"), dimethylformamide ("DMF"), chloroform, methylene chloride (or dichloromethane), diethyl ether, methanol, pyridine and the like]. Unless specified to the contrary, the solvents used in the reactions of the present invention are inert organic solvents.

[0052] The term "q.s." means adding a quantity sufficient to achieve a stated function, e.g., to bring a solution to the desired volume (i.e., 100%).

[0053] In general, esters of carboxylic acids may be prepared by conventional esterification procedures, for example alkyl esters may be prepared by treating the required carboxylic acid with the appropriate alkanol, generally under acidic conditions. Likewise, amides may be prepared using conventional amidation procedures, for example amides may be prepared by treating an activated carboxylic acid with the appropriate amine. Alternatively, a lower-alkyl ester such as a methyl ester of the acid may be treated with an amine to provide the required amide, optionally in presence of trimethylaluminum following the procedure described in Tetrahedron Lett. 48, 4171-4173, (1977). Carboxyl groups may be protected as alkyl esters, for example methyl esters, which esters may be prepared and removed using conventional procedures, one convenient method for converting carbomethoxy to carboxyl is to use aqueous lithium hydroxide.

[0054] The salts and solvates of the compounds mentioned herein may as required be produced by methods conventional in the art. For example, if an inventive compound is an acid, a desired base addition salt can be prepared by treatment of the free acid with an inorganic or organic base, such as an amine (primary, secondary, or tertiary); an alkali metal or alkaline earth metal hydroxide; or the like. Illustrative examples of suitable salts include organic salts derived from amino acids such as glycine and arginine; ammonia; primary, secondary, and tertiary amines; such as ethylenediamine, and cyclic amines, such as cyclohexylamine, piperidine, morpholine, and piperazine; as well as inorganic salts derived from sodium, calcium, potassium, magnesium, manganese, iron, copper, zinc, aluminum, and lithium.

[0055] If a compound is a base, a desired acid addition salt may be prepared by any suitable method known in the art, including treatment of the free base with an inorganic acid,

such as hydrochloric acid, hydrobromic acid, sulfuric acid, nitric acid, phosphoric acid, and the like, or with an organic acid, such as acetic acid, maleic acid, succinic acid, mandelic acid, fumaric acid, malonic acid, pyruvic acid, oxalic acid, glycolic acid, salicylic acid, pyranosidyl acid, such as glucuronic acid or galacturonic acid, alpha-hydroxy acid, such as citric acid or tartaric acid, amino acid, such as aspartic acid or glutamic acid, aromatic acid, such as benzoic acid or cinnamic acid, sulfonic acid, such as p-toluenesulfonic acid, methanesulfonic acid, ethanesulfonic acid, or the like.

[0056] Isolation and purification of the compounds and intermediates described herein can be effected, if desired, by any suitable separation or purification procedure such as, for example, filtration, extraction, crystallization, column chromatography, thin-layer chromatography or thick-layer chromatography, or a combination of these procedures. Specific illustrations of suitable separation and isolation procedures can be had by reference to the examples hereinbelow. However, other equivalent separation or isolation procedures can, of course, also be used.

Synthesis of the Compounds of Formula I

[0057] The compounds of Formula I can be prepared by following the procedures described with reference to the Reaction Schemes below. See, also, See, e.g., Gudmundsson, K. S., Hinkley, J. M., Brieger, M. S., Drach, J. C., Townsend, L. B. *Synthetic Communications*, 27(5), 861-870, 1997; PCT Publication Nos. WO 01/30768, WO 01/98278, WO 03/39460, WO 03/49678, WO 03/50122, WO 03/49527, WO 0349679, and WO 03/50064, each of which is incorporated herein by reference for all purposes.

Brief Description of Reaction Schemes

[0058] Reaction Scheme 1 illustrates synthesis of compounds of Formula I wherein R_3 is $-\text{COR}_6$.

[0059] Reaction Scheme 2 illustrates a synthesis of compounds of Formula I wherein R_3 is $-\text{SO}_2\text{R}_{6a}$.

[0060] Reaction Scheme 3 illustrates a synthesis of compounds of Formula I.

[0061] Reaction Scheme 4 illustrates a synthesis of compounds of Formula I wherein R_3 taken together with R_7 form an optionally substituted imidazolyl.

[0062] Reaction Scheme 5 illustrates an alternative synthesis of compounds of

Formula I wherein R₃ taken together with R₇ form an optionally substituted imidazolyl.

[0063] Reaction Scheme 6 illustrates a synthesis of compounds of Formula I wherein R₃ taken together with R₇ form an optionally substituted imidazolyl.

[0064] Reaction Scheme 7 illustrates an alternative synthesis of compounds of Formula I wherein R₃ taken together with R₇ form an optionally substituted imidazolyl.

[0065] Reaction Scheme 8 illustrates a synthesis of compounds of Formula I wherein R₃ is -COR₆ and R₆ is -OR₁₁.

[0066] Reaction Scheme 9 illustrates a synthesis of compounds of Formula I wherein R₃ is -COR₆ and R₆ is -NHR₁₂.

[0067] Reaction Scheme 10 illustrates a synthesis of compounds of Formula I wherein R₃ taken together with R₇ form an optionally substituted diazepinone.

[0068] Reaction Scheme 11 illustrates a synthesis of compounds of Formula I wherein R₃ taken together with R₇ form an optionally substituted diazepinone.

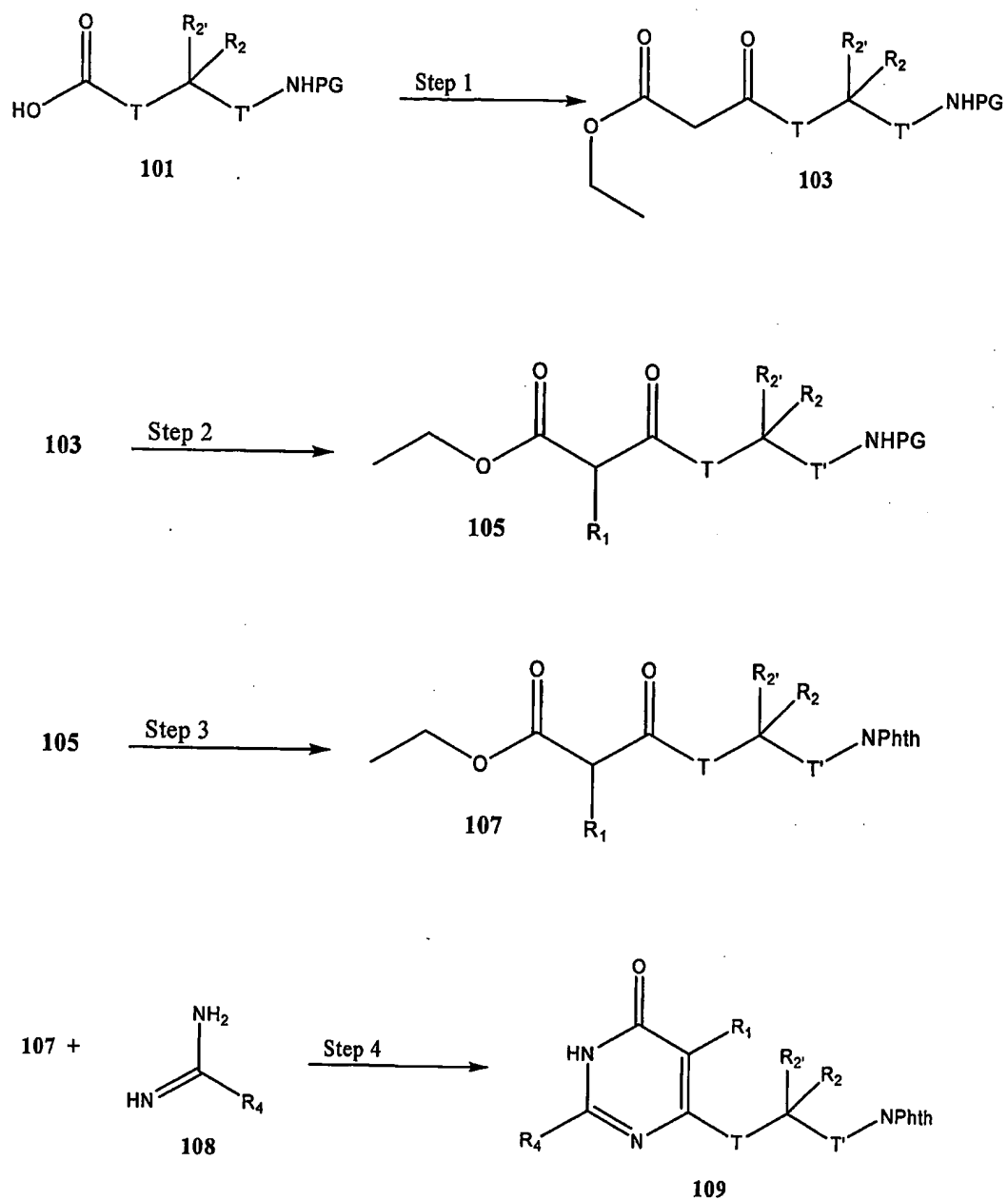
[0069] Reaction Scheme 12 illustrates a synthesis of compounds of Formula I wherein R₃ taken together with R₇ form an optionally substituted heterocyclic ring.

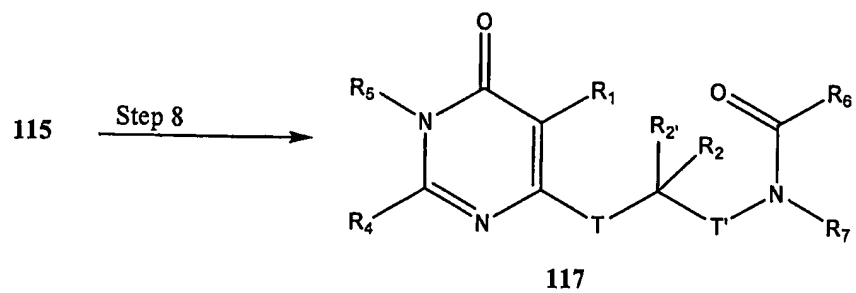
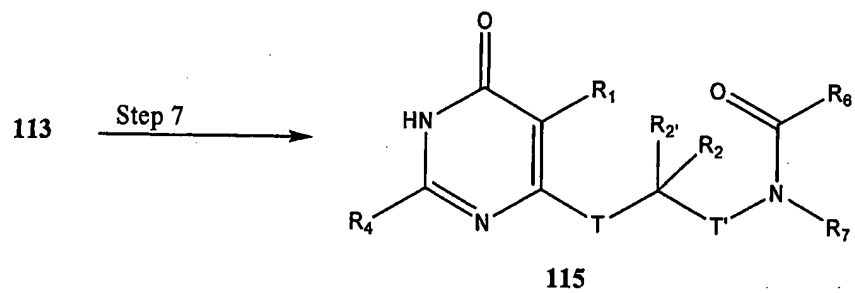
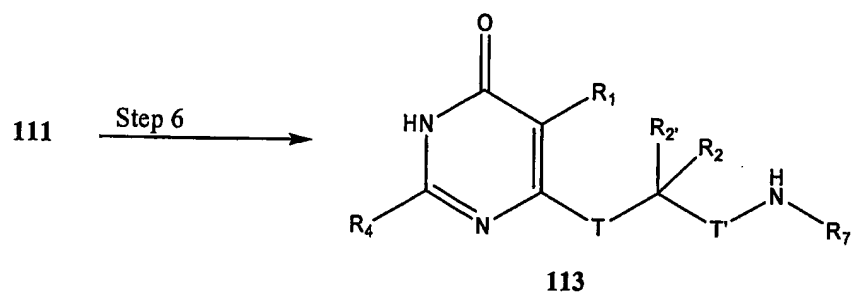
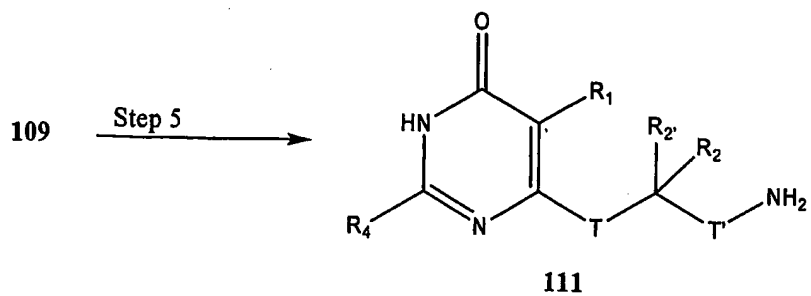
[0070] Reaction Scheme 13 illustrates a synthesis of compounds of Formula I wherein R₄ and R₅ taken together form an optionally substituted nitrogen-containing heterocycle,

Starting Materials

[0071] The optionally substituted compounds of Formula 101 and other reactants are commercially available, e.g., from Aldrich Chemical Company, Milwaukee, WI, or may be readily prepared by those skilled in the art using commonly employed synthetic methodology.

Reaction Scheme 1





Preparation of Compounds of Formula 103

[0072] Referring to Reaction Scheme 1, Step 1, a solution of lithium ethyl acetate is first produced. To a solution of an excess, preferably about 3.3 equivalents, of diisopropylamine in an anhydrous, nonpolar, aprotic solvent such as tetrahydrofuran at about 0°C is added an excess, preferably about 3.3 equivalents, of n-butyllithium (preferably as a 2.5 M solution in THF) and the resulting mixture is stirred at about the same temperature for about another hour. After this mixture is cooled to about -78°C, about three equivalents of ethyl acetate is added, and the resulting solution is stirred at about the same temperature for about 1.5 hours.

[0073] Meanwhile, to a solution of one equivalent of an N-protected amino acid of Formula 101 (preferably wherein the amino protecting group, PG, is a Boc group) in an anhydrous, nonpolar, aprotic solvent such as tetrahydrofuran at room temperature is added about one equivalent of 1,1'-carbonyldiimidazole. The resulting mixture is stirred at room temperature for about 1.5 hours. This mixture is added to the above freshly prepared lithium ethyl acetate solution at about -78°C. After being stirred for about 2 hours at about the same temperature, the product, a compound of Formula 103, is isolated and purified.

Preparation of Compounds of Formula 105

[0074] Referring to Reaction Scheme 1, Step 2, to a mixture of a compound of Formula 103 and a base such as potassium carbonate in acetone at room temperature is added an excess, preferably about 1.1 equivalents, of the formula R_1X wherein X is chloro, bromo, iodo, tosylate or another suitable leaving group, as will be readily appreciated by one of skill in the art. The resulting solution is heated to about 60°C for about 4 hours until TLC indicates that almost no starting material is present. The product, a compound of Formula 105, is isolated as a mixture of diastereomers and purified.

Preparation of Compounds of Formula 107

[0075] Referring to Reaction Scheme 1, Step 3, if PG is a Boc protecting group, it is removed in favor of a phthalate group. Accordingly to a solution of a compound of Formula 105 in a nonpolar, aprotic solvent such as dichloromethane at about 0 °C is added trifluoroacetic acid. The resulting solution is stirred at room temperature for about 2 hours and then concentrated under reduced pressure. The residue is dried under high vacuum and used in the next step without further purification.

[0076] To a solution of the above crude residue in a nonopolar, aprotic solvent such

as dioxane is added an excess, preferably about 3 equivalents, of phthalic anhydride and the resulting solution is heated to about 100 °C until LC/MS indicated almost no starting material is present. The product, a compound of Formula 107, is isolated as a mixture of diastereoisomers and purified.

Preparation of Compounds of Formula 109

[0077] Referring to Reaction Scheme 1, Step 4, to a solution of an excess, preferably about 2 equivalents, of an ester of Formula 107 in a polar, protic solvent such as methanol is added an amidine of Formula 108 and a strong base such as sodium methoxide (preferably about 0.6 equivalent of a 0.5 M solution of sodium methoxide in methanol) successively at room temperature. The resulting solution is heated to about 60 °C for about 48 hours. The pyrimidinone of Formula 109 is isolated and purified.

Preparation of Compounds of Formula 111

[0078] Referring to Reaction Scheme 1, Step 5, the phthalate group is removed. To a solution of a pyrimidinone of Formula 109 in acetic acid at room temperature is added aqueous hydrochloric acid. The resulting solution is stirred at about 110 °C for about 1 hour and monitored by LC/MS. The product, a compound of Formula 111 is isolated and purified.

Preparation of Compounds of Formula 113

[0079] Referring to Reaction Scheme 1, Step 6, to a solution of a pyrimidinone of Formula 111 in a nonpolar, aprotic solvent such as tetrahydrofuran at room temperature are added an excess, preferably about 1.2 equivalents, of sodium triacetoxyborohydride and an excess, preferably about 1.4 equivalents, of an aldehyde of formula $R_7\text{-CHO}$ (preferably wherein R_7 includes a protected amino alkyl group and wherein $R_7\text{-CH}_2\text{-}$ corresponds to R_7) successively. The resulting mixture is stirred at room temperature under a nitrogen atmosphere for about 4 hours until almost no starting material is present. The product, a compound of Formula 113, is isolated and purified.

Preparation of Compounds of Formula 115

[0080] Referring to Reaction Scheme 1, Step 7, to a solution of a pyrimidinone of Formula 113 in a nonpolar, aprotic solvent such as dichloromethane at about 0 °C are added

an amine base such as diisopropylethylamine and an excess, preferably about three equivalents, of an acid chloride of formula $R_6(CO)Cl$, successively. The resulting solution is stirred at room temperature under a nitrogen atmosphere overnight. The solvents are evaporated and the residue is dissolved in a protic, polar solvent such as methanol. Aqueous base (preferably, aqueous potassium carbonate) is added until the pH is about 8. The resulting solution is stirred at room temperature for about an hour. The product, a compound of Formula 115, is isolated and purified.

Preparation of Compounds of Formula 117

[0081] Referring to Reaction Scheme 1, Step 8, to a solution of a compound of Formula 115, in a nonpolar, aprotic solvent, such as dioxane is added an excess, preferably about three equivalents, of lithium hydride and an excess, preferably about 1.5 equivalents, of dimethyl sulfate, successively. The resulting mixture is stirred at room temperature for about 4 hours and at about 40 °C for about 15 minutes. The product, a compound of Formula 117, is isolated and used without further purification.

[0082] Optionally, any protecting groups can then be removed. For example, if R_7 is a protected aminoalkyl- group, the amine protected group can be removed to yield the free amine. In an embodiment when a Boc group is used, it can be removed by treating a compound of Formula 115 in a nonpolar, aprotic solvent such as dichloromethane at about 0 °C with trifluoroacetic acid. The resulting solution is stirred at room temperature for about 2 hours and then concentrated under reduced pressure. The product is isolated and purified.

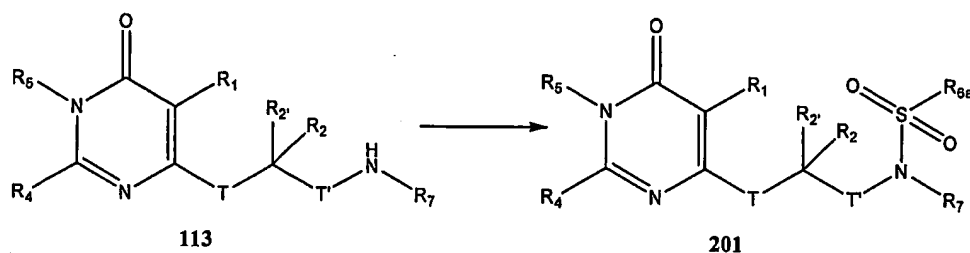
Preparation of Optically Active Compounds

[0083] In certain compounds of the invention, a particular stereo configuration (such as the (R) isomer) may be preferred at the stereogenic center to which R_2 is attached. The optically active compound can be prepared by methods known in the art. For example, an amine of Formula 111 is dissolved in an inert organic solvent (such as IPA) and warmed to 60°C. In a separate vessel, a resolving agent (such as dibenzoyl-D-tartaric acid) is dissolved, preferably in the same warm solvent, and then quickly added (with agitation) to the warm amine solution. The reaction mixture is left to crystallize by cooling to room temperature over 16 hours under continuing agitation. The desired isomer, e.g., the (R) isomer, is isolated and purified.

[0084] For the sake of brevity in the remaining description of the synthesis of

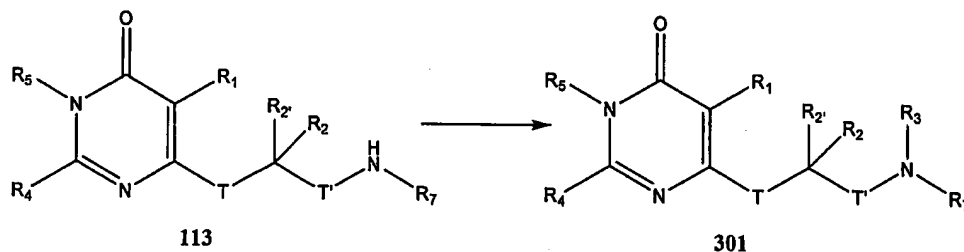
compounds of Formula I, it should be understood that either single isomer or a mixture of isomers may be employed to give the corresponding product.

Reaction Scheme 2



[0085] Referring to Reaction Scheme 2, to a solution of a compound of Formula 113 and an amine base (such as diisopropylethylamine) in a polar, aprotic solvent (such as dichloromethane) is added a compound having the formula $Cl-S(O)_2-R_{6a}$ or $O-(S(O)_2-R_{6a})_2$ where R_{6a} is as described above. The resulting solution is stirred under nitrogen at room temperature for several hours. The product, a compound of Formula 201 is isolated and purified.

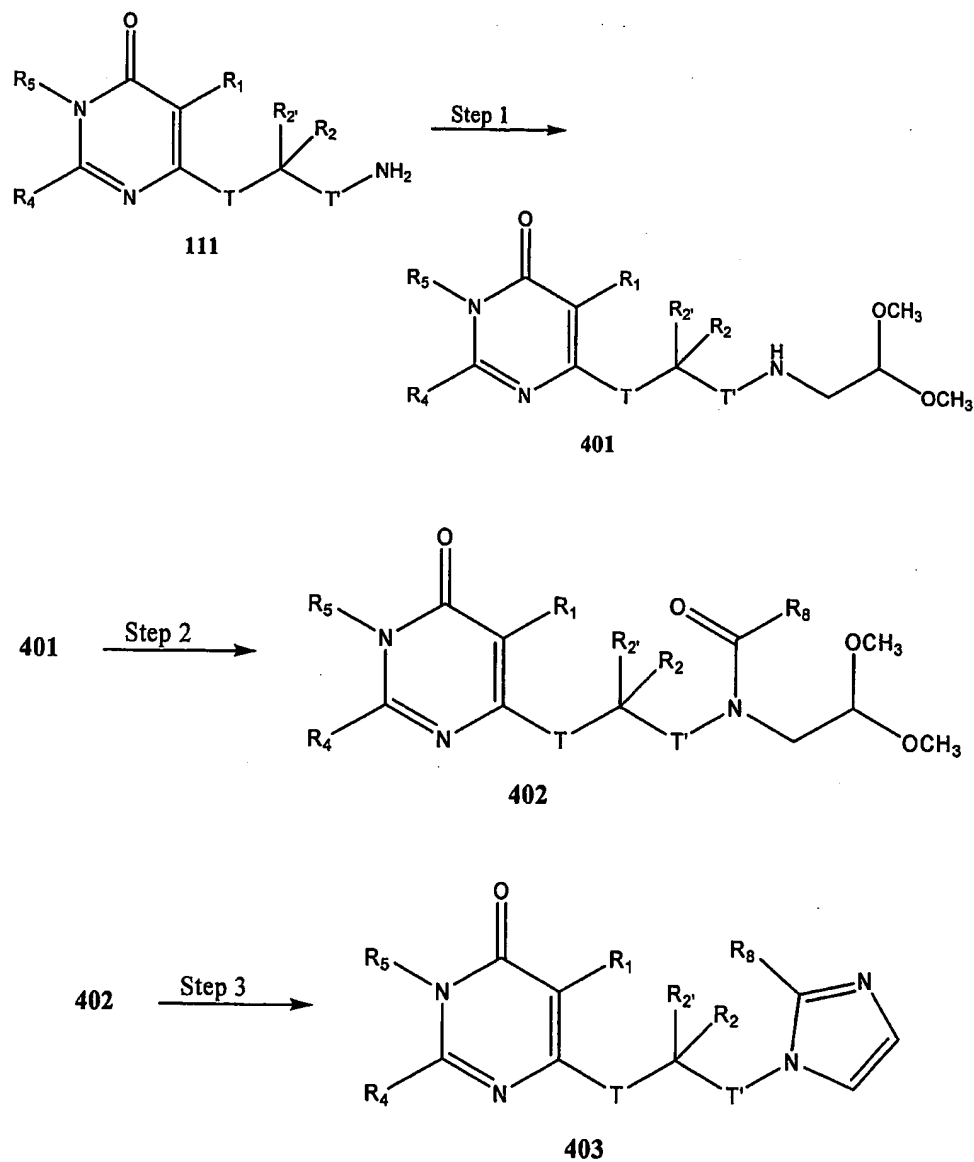
Reaction Scheme 3



[0086] Referring to Reaction Scheme 3, to a solution of a compound of Formula 113 and an amine base (such as diisopropylethylamine) in a polar, aprotic solvent (such as dichloromethane) is added a compound having the formula $X-R_3$ where X is a leaving group and especially bromo or chloro and R_3 is as described above. The resulting solution is stirred under nitrogen at room temperature or with heat for several hours. The product, a compound

of Formula 301 is isolated and purified.

Reaction Scheme 4



Preparation of Formula 401

[0087] Referring to Reaction Scheme 4, Step 1, to an optionally substituted compound of Formula 111 dissolved in a polar, aprotic solvent (such as DMF) in the presence

of a base (such as potassium carbonate) is added one equivalent of an optionally substituted suitably protected aldehyde wherein such aldehyde further comprises a leaving group, preferably, a halide. The solution is heated at reflux, monitoring completion of the reaction (e.g., by TLC). The reaction mixture is cooled and the corresponding, optionally substituted compound of Formula 401 is isolated and purified.

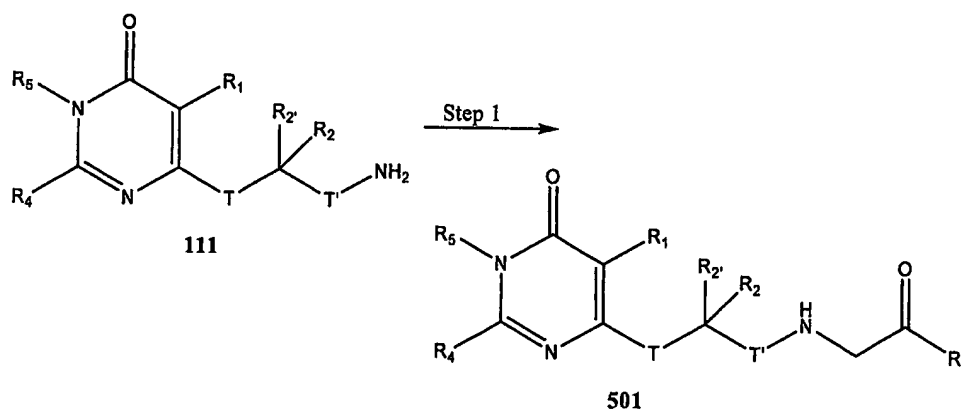
Preparation of Formula 402

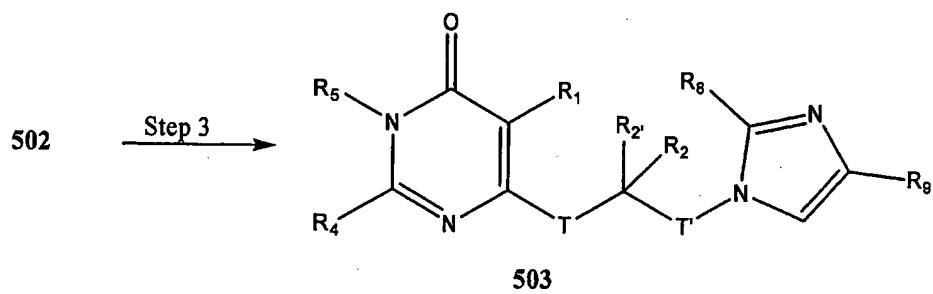
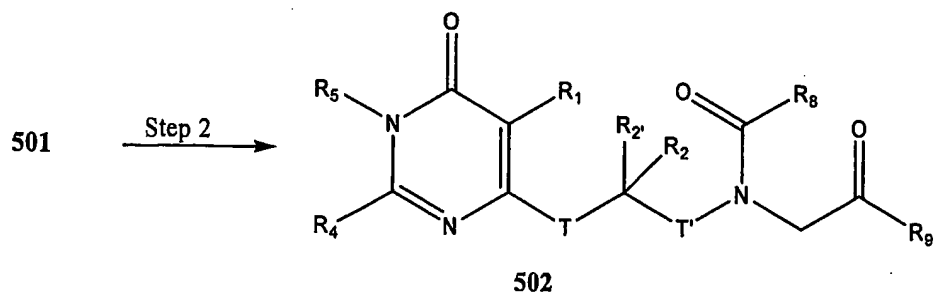
[0088] Referring to Reaction Scheme 4, Step 2, to an optionally substituted compound of Formula 402 in an inert solvent (such as dichloromethane) in the presence of about 1.5 molar equivalents of an amine base (such as triethylamine) is added about 1.5 molar equivalents of an R_8 acid chloride, such as, $Cl-C(O)-R_8$, where R_8 is as described below. The reaction takes place, with stirring, at room temperature over a period of 4 to 24 hours. Completion is monitored, e.g., by TLC. The corresponding compound of Formula 402 is isolated and purified.

Preparation of Formula 403

[0089] Referring to Reaction Scheme 4, Step 3, a solution of a compound of Formula 402 and an excess of ammonium acetate in acetic acid is heated at reflux for 1-4 hours. Completion is monitored, e.g., by TLC. The corresponding compound of Formula 403 is isolated and purified.

Reaction Scheme 5





Preparation of Formula 501

[0090] Referring to Reaction Scheme 5, Step 1, a suspension of a compound of Formula 111, an alpha-haloketone reagent of the Formula $R_9(CO)CH_2X$ wherein X is a halide, and about an equivalent of a base, such as potassium carbonate in a polar, aprotic solvent such as DMF is stirred at room temperature. The reaction is diluted with water and the resulting solid, a compound of Formula 501, is used in the subsequent step without further purification.

Preparation of Formula 502

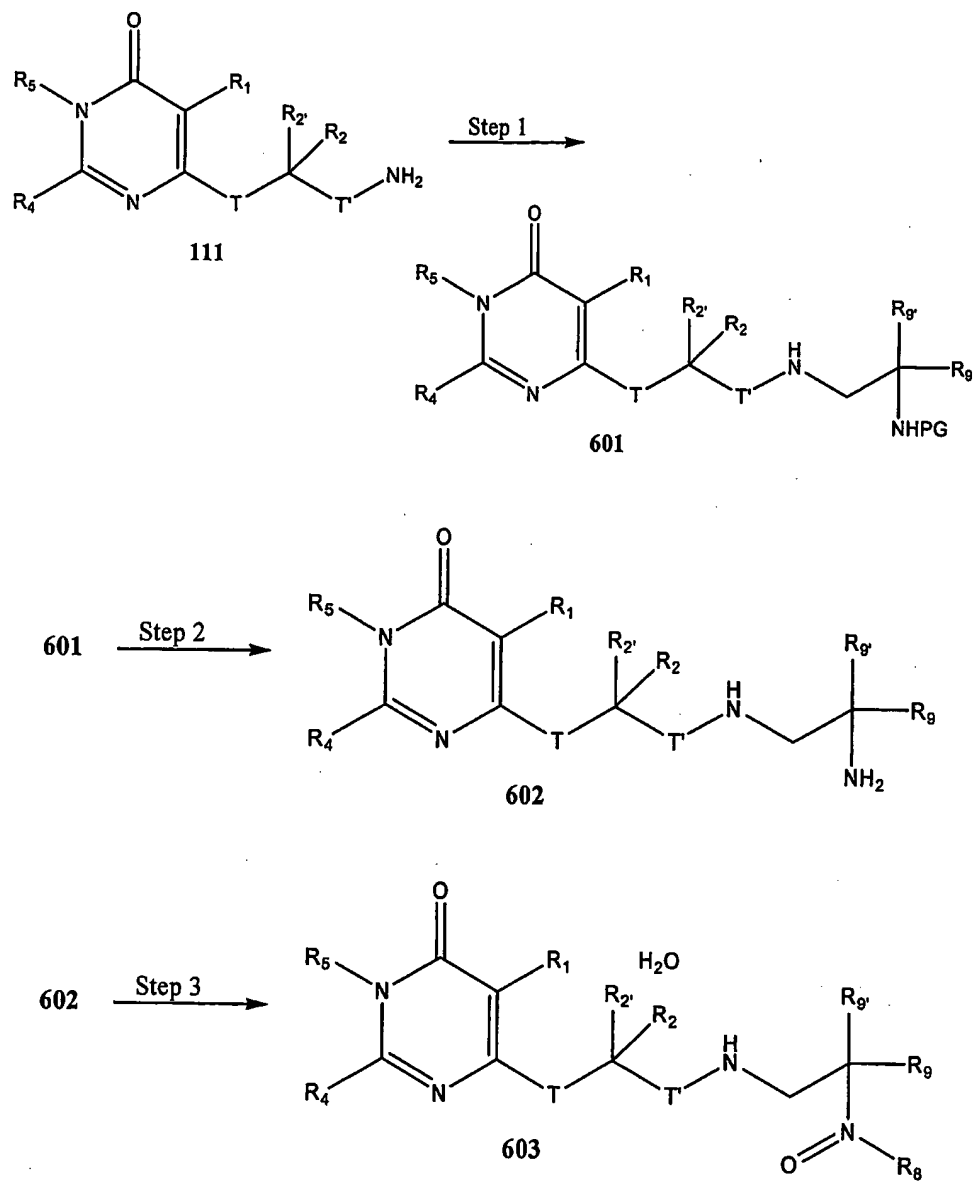
[0091] Referring to Reaction Scheme 5, Step 2, a solution of the compound of Formula 501, about an equivalent of an amine base, such as triethylamine and about an equivalent of an acid chloride (such as a compound of Formula R_8-COCl) in an organic solvent such as methylene chloride is stirred at room temperature for several hours. Completion is monitored, e.g., by TLC. The corresponding compound of Formula 502 is isolated and purified.

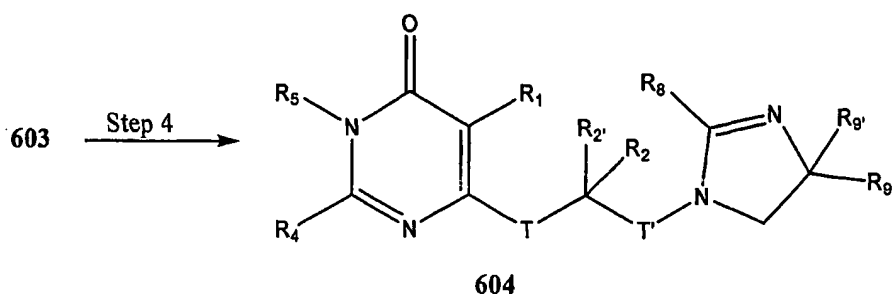
Preparation of Formula 503

[0092] Referring to Reaction Scheme 5, Step 3, a solution of a compound of Formula 502 and an excess of ammonium acetate in acetic acid is heated at reflux using a Dean-Stark trap and condenser. Completion is monitored, e.g., by TLC. The corresponding compound of Formula 503 is isolated and purified.

[0093] Optionally, any protecting groups on compounds of Formula 507 are then removed. For example, if a phthalimide protecting group is used, it may be removed as follows. A solution of a compound of Formula 503 and an excess of anhydrous hydrazine in a polar, protic solvent such as ethanol is heated at reflux. The reaction is cooled to about 50°C and any precipitate is filtered off. The filtrate is concentrated in vacuo and purified to yield the corresponding free amine. One of skill in the art will appreciate that other conditions may be used to remove other protecting groups.

Reaction Scheme 6





Preparation of Formula 601

[0094] Referring to Reaction Scheme 6, Step 1, reductive amination of amines of Formula 111 with an optionally substituted, aldehyde-containing carbamic acid ester gives urethane intermediates. Removal of the Boc protecting group furnishes an amine of Formula 602.

[0095] More specifically, to a solution of a compound of Formula 111 and an equivalent of a suitably protected aldehyde (Seki *et. al. Chem. Pharm. Bull.* 1996, 44, 2061) in dichloromethane is added a slight excess of a reducing agent, such as sodium triacetoxyborohydride. The resultant cloudy mixture is maintained at ambient temperature. Completion is monitored, e.g., by TLC. The corresponding compound of Formula 601 is isolated and used in the subsequent step without purification.

Preparation of Formula 602

[0096] Referring to Reaction Scheme 6, Step 2, to a solution of a compound of Formula 601 in a polar, aprotic solvent such as dichloromethane is added a strong acid such as trifluoroacetic acid. The resultant solution is maintained at ambient temperature overnight and concentrated under reduced pressure. The residue is isolated to give a compound of Formula 602 which is used in the subsequent step without purification.

Preparation of Formula 603

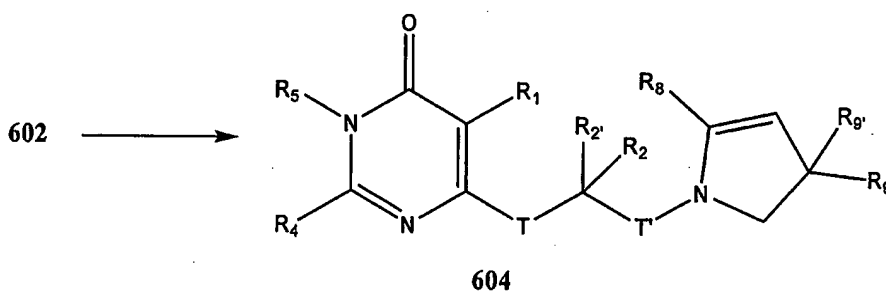
[0097] Referring to Reaction Scheme 6, Step 3, to a solution of a compound of Formula 602 in a polar, aprotic solvent such as dichloromethane is added an excess, preferably about two equivalents of an amine base such as triethylamine, followed by about an equivalent or slight excess of an acid chloride. The resultant solution is stirred at ambient

temperature for about 3 hours. Completion is monitored, e.g., by TLC. The corresponding compound of Formula 603 is isolated and purified.

Preparation of Formula 604

[0098] Referring to Reaction Scheme 6, Step 4, a solution of a compound of Formula 603 in an excess of phosphorus oxychloride is heated at reflux. After 8 hours, the reaction mixture is allowed to cool to ambient temperature and concentrated under reduced pressure. The corresponding compound of Formula 604 is isolated and purified.

Reaction Scheme 7

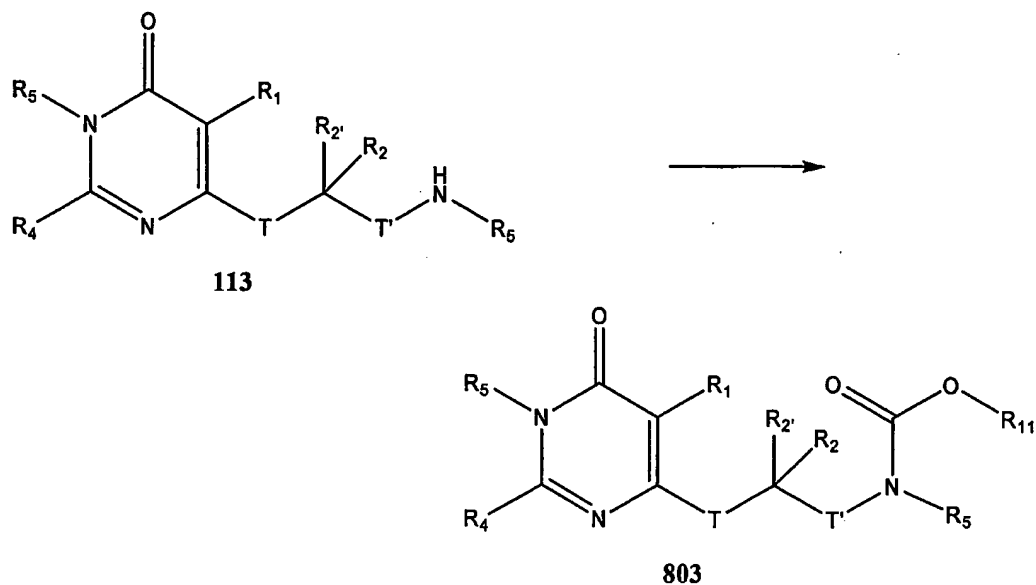


Preparation of Formula 604

[0099] As an alternative to Steps 3 and 4 of Reaction Scheme 6, acylation of primary amines of Formula 602, followed by acetic acid mediated cyclization, can proceed without isolation of the intermediate amides to provide the target compound of Formula 604. This route is shown in Reaction Scheme 6.

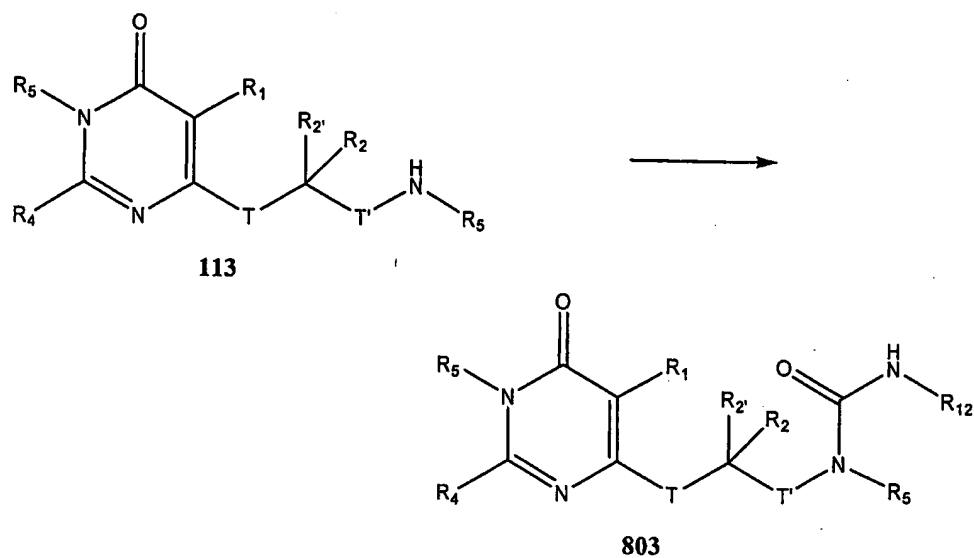
[00100] More specifically, to a solution of a compound of Formula 602 in a polar, aprotic solvent such as dichloromethane is added an excess, preferably about two equivalents of an amine base, such as triethylamine, followed by about an equivalent of an acid chloride. The resultant solution is stirred at ambient temperature for 2 hours, then evaporated under reduced pressure. The resultant solid is treated with glacial acetic acid, then the resultant suspension is heated at reflux for about 48 hours. The reaction is cooled to ambient temperature then evaporated under reduced pressure. The corresponding compound of Formula 604 is isolated and purified.

Reaction Scheme 8



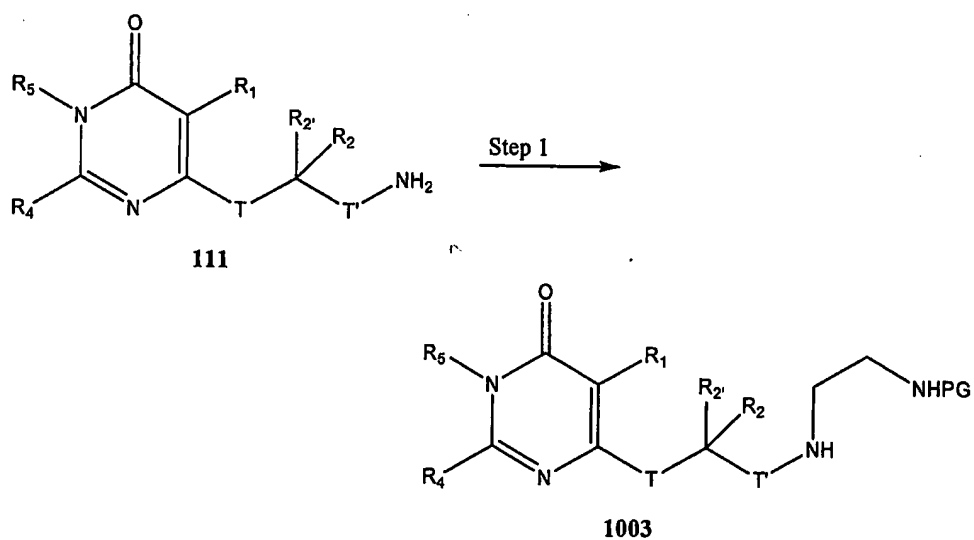
[00101] Referring to Reaction Scheme 8, a compound of Formula 113 is reacted with a slight excess of a compound of the formula $R_{11}O(CO)Cl$ in the presence of a base such as triethylamine in a nonpolar, aprotic solvent such as dichloromethane. The product, a compound of Formula 803 is isolated and purified.

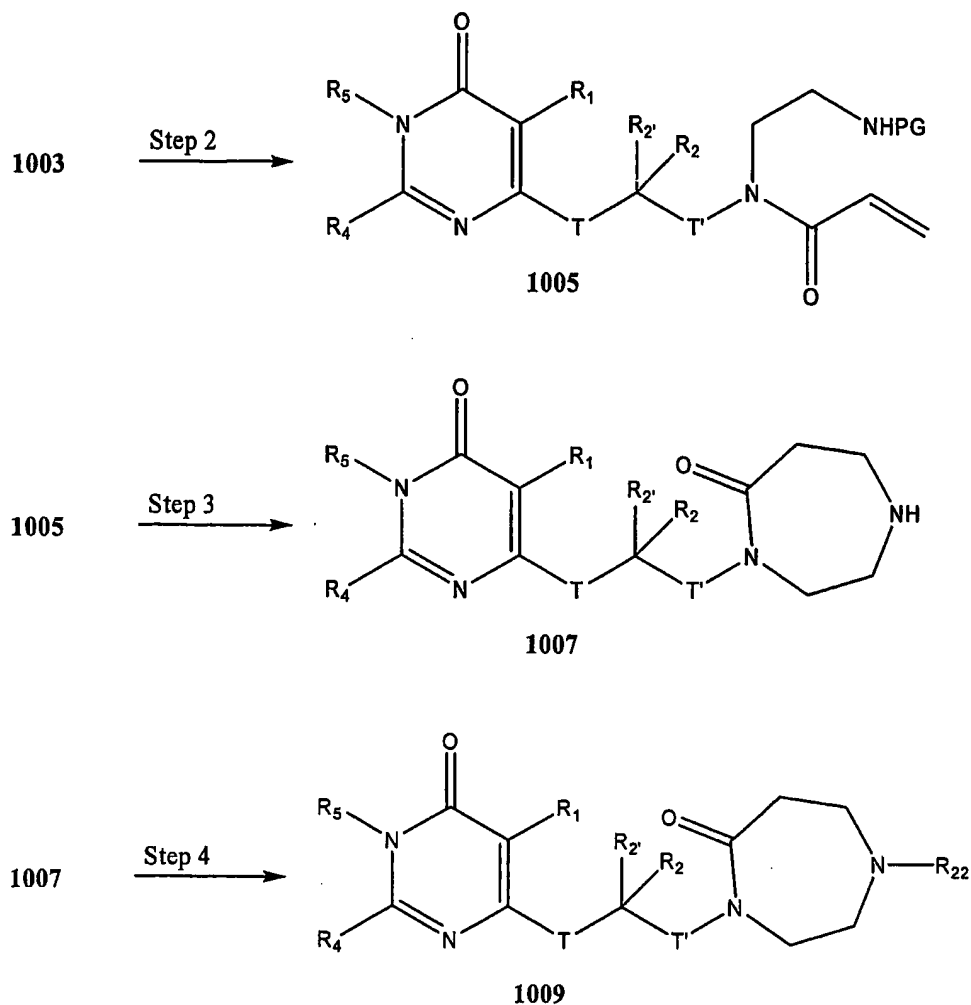
Reaction Scheme 9



[00102] Referring to Reaction Scheme 9, a compound of Formula 113 is treated with a slight excess of an isocyanate $R_{12}-N=C=O$ in the presence of a base, such as triethylamine, in a nonpolar, aprotic solvent, such as dichloromethane. The product, a compound of Formula 903, is isolated and purified.

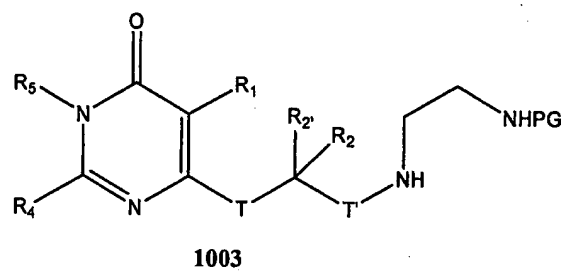
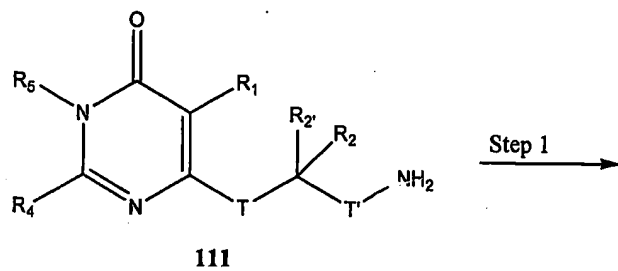
Reaction Scheme 10





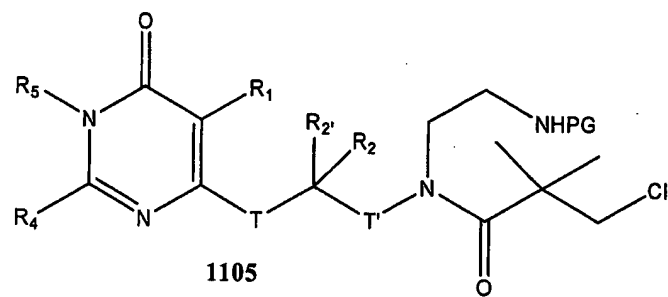
[00103] Referring to Reaction Scheme 10, reductive amination of the primary amino group in compounds of Formula 111 with (2-oxo-ethyl)-carbamic acid *tert*-butyl ester gave the corresponding secondary amine. Acylation with acryloyl chloride followed by deprotection of the tertiary amide and base mediated cyclisation gave the desired diazepanones. If desired, further functionalization of the basic amine could be accomplished under conditions well known to those skilled in the art.

Reaction Scheme 11



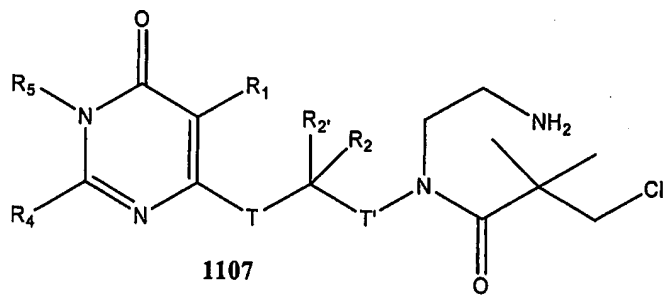
1003

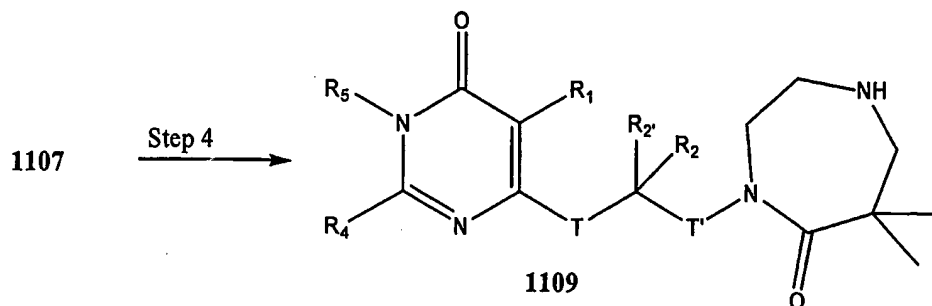
Step 2



1105

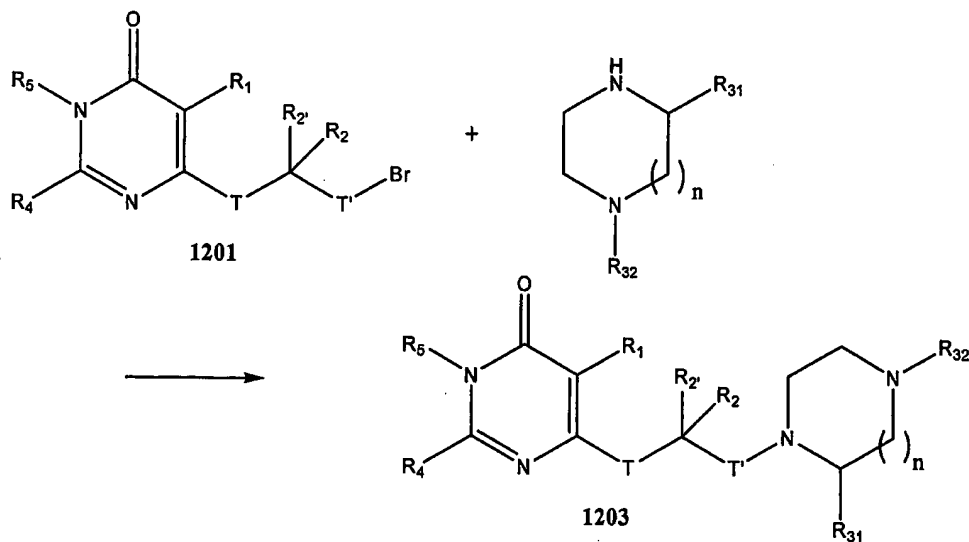
Step 3





[00104] Referring to Reaction Scheme 11, reductive amination of the primary amino group in compounds of Formula 111 with (2-oxo-ethyl)-carbamic acid *tert*-butyl ester gave the corresponding secondary amine. Acylation with chloropivaloyl chloride followed by deprotection of the tertiary amide and base mediated cyclisation gave the desired diazepanones. If desired, further functionalization of the basic amine could be accomplished under conditions well known to those skilled in the art.

Reaction Scheme 12

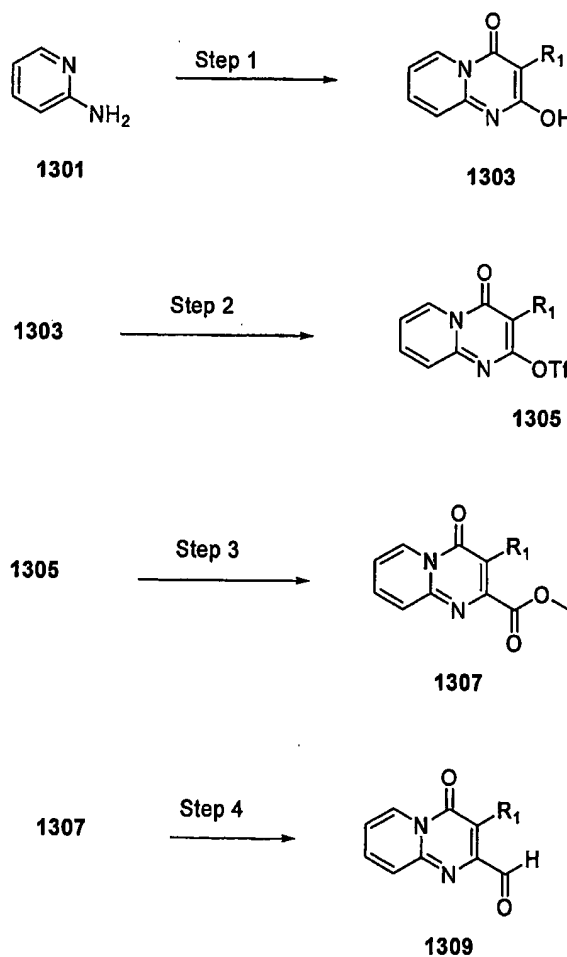


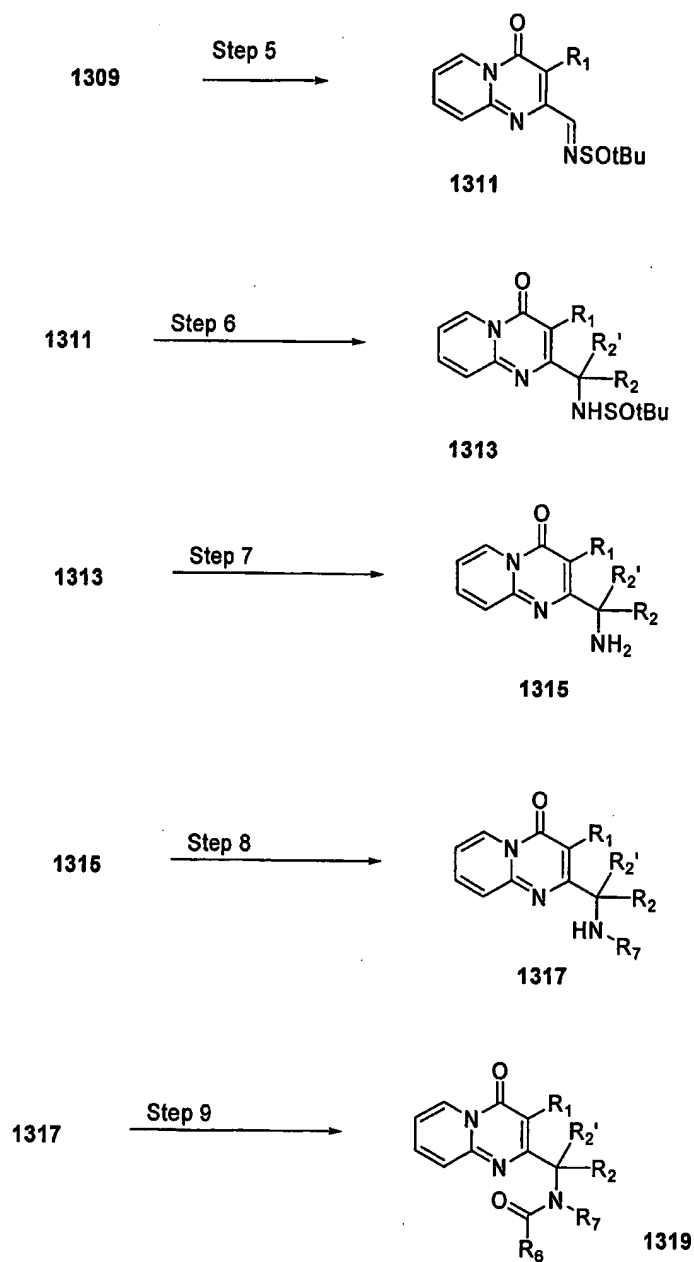
[00105] Referring to Reaction Scheme 12, a compound of Formula 1201, one-half molar equivalent of an optionally substituted piperazine or diazepam (as shown above, where R_{32} is as described herein) and an excess of potassium carbonate are combined in an organic solvent (e.g., acetonitrile). The reaction takes place under a nitrogen atmosphere at elevated temperature (e.g., 100°C) over a period of 8 hours, followed at a somewhat lower temperature

(e.g., 60°C) for a period of 5 days. The product, a compound of Formula 1203, is isolated and purified.

[00106] Optionally, in the event that R₃₂ is an amine protecting group, such as Boc, it may be removed by for example treatment with a 95/5 mixture of TFA/water followed by stirring at room temperature for 1 hour. The product, a compound of Formula 1203 wherein R₃₂ is hydrogen, can be isolated and purified. If desired, further functionalization of the basic amine could be accomplished under conditions well known to those skilled in the art.

Reaction Scheme 13



**Preparation of Compounds of Formula 1303**

[00107] Referring to Reaction Scheme 13, Step 1, a solution of an optionally substituted

aminopyridine of Formula 1301 and a malonic acid of the formula $R_1\text{-CH}(\text{CO}_2\text{H})_2$, and POCl_3 is maintained at about 100°C for about 2 hours. One of skill in the art will appreciate that the aminopyridine of Formula 1301 can be substituted with a variety of groups on the aromatic ring, as described below. The solution is then concentrated and to the resulting residue is added ethyl acetate and ice water. The resulting mixture is stirred vigorously for about 30 mins at about 0°C . The product, a compound of Formula 1303, is isolated and used without further purification.

Preparation of Compounds of Formula 1305

[00108] Referring to Reaction Scheme 13, Step 2, the ring hydroxyl group of a compound of Formula 1303 is converted to the corresponding triflate. Accordingly, a solution of pyridopyrimidinone of Formula 1303, an excess (preferably about two equivalents) of N-phenylbis(trifluoromethane-sulfonamide), a base such as triethylamine, and a polar, aprotic solvent such as DMF is maintained at about 50°C for about 30 mins. The product, a compound of Formula 1305, is isolated and used without further purification.

Preparation of Compounds of Formula 1307

[00109] Referring to Reaction Scheme 13, Step 3, a Parr bottle containing a solution of crude pyridopyrimidinone triflate of formula 1305, about 0.1 equivalent of $\text{Pd}(\text{OAc})_2$, about 0.1 equivalent of 1,3-bis(diphenylphosphino)propane, a base such as triethylamine, and a polar, protic solvent such as MeOH is purged with carbon monoxide for about 10 minutes. The apparatus is sealed, pressurized with carbon monoxide (to about 50 psi), and heated to 70°C for about 2 hours. The product, a compound of Formula 1307, is isolated and purified.

Preparation of Compounds of Formula 1309

[00110] Referring to Reaction Scheme 13, Step 4, an excess (preferably about 2 equivalents) of diisobutylaluminum hydride (preferably, a 1.5 M in toluene) is added dropwise to an about -78°C solution of pyridopyrimidinone ester of Formula 107 and a nonpolar, aprotic solvent such as toluene. After about 15 mins, the product, a compound of Formula 1309, is isolated and purified.

Preparation of Compounds of Formula 1311

[00111] Referring to Reaction Scheme 13, Step 5, a mixture of pyridopyrimidinone

aldehyde of Formula 109, an excess (preferably about 1.6 equivalents) of (S)-(-)-2-methyl-2-propanesulfonamide, about 0.1 equivalent of pyridinium *p*-toluenesulfonate, anhydrous magnesium sulfate, and a nonpolar, aprotic solvent such as CH₂Cl₂ is maintained at room temperature for about 18 hours. The product, a compound of Formula 1311, is isolated and used without additional purification.

Preparation of Compounds of Formula 1313

[00112] Referring to Reaction Scheme 13, Step 6, isopropyl magnesium bromide (preferably as a 1.0 M solution in THF) is added dropwise to a -78°C solution of a compound of Formula 1311 and a nonpolar, aprotic solvent such as THF. After about 30 mins, the product, a compound of Formula 1313, is isolated and purified.

Preparation of Compounds of Formula 1315

[00113] Referring to Reaction Scheme 13, Step 7, a solution of pyridopyrimidine of Formula 1313, a polar, protic solvent such as MeOH, and an acid (preferably 4 N HCl in dioxane) is maintained at room temperature for about 30 mins. The product, a compound of Formula 1315, is isolated and used without further purification.

Preparation of Formula 1317

[00114] Referring to Reaction Scheme 13, Step 8, to a solution of a compound of Formula 1315 is added successively a slight excess (preferably about 1.2 equivalents) of an aldehyde comprising R₇ (i.e., a compound having the formula R₇-CHO where R₇-CH₂- is equivalent to R₇ and R₄ is as described above or is a protected precursor to such a substituent, e.g., (3-oxo-propyl)-carbamic acid *tert*-butyl ester) and a reducing agent such as sodium triacetoxyborohydride. The resulting mixture is stirred for several hours. The product, a compound of Formula 1317 is isolated and purified.

Preparation of Formula 1319

[00115] Referring to Reaction Scheme 13, Step 9, to a solution of a compound of Formula 1317 and an amine base such as diisopropylethylamine in a nonpolar, aprotic solvent such as dichloromethane is added an R₆ acyl chloride (such as Cl-C(O)-R₆ where R₆ is as described above). The resulting solution is stirred under nitrogen at room temperature for several hours. The product, a compound of Formula 1319 is isolated and purified.

[00116] Optionally, any protecting groups on compounds of Formula 1319 are then removed. For example, if R₇ comprises a protected amine wherein the protecting group is a Boc group, the Boc group can be removed by treatment of the compound of Formula 1319 with an acid such as trifluoroacetic acid in a nonpolar, aprotic solvent such as dichloromethane, while maintaining the reaction at about room temperature. The reaction is monitored e.g., by TLC. Upon completion, the corresponding free amine is isolated and purified.

Preferred Processes and Last Steps

[00117] A compound of Formula I is contacted with a pharmaceutically acceptable acid to form the corresponding acid addition salt.

[00118] A pharmaceutically acceptable acid addition salt of Formula I is contacted with a base to form the corresponding free base of Formula I.

[00119] A compound of Formula 115 wherein R₇ is a Boc protected aminoalkyl group is deprotected by contact with an acid such as trifluoroacetic acid to give the corresponding free amine.

[00120] A compound of Formula 117 wherein R₇ is a Boc protected aminoalkyl group is deprotected by contact with an acid such as trifluoroacetic acid to give the corresponding free amine.

[00121] A compound of Formula 1319 wherein R₇ is a Boc protected aminoalkyl group is deprotected by contact with an acid such as trifluoroacetic acid to give the corresponding free amine.

Preferred Compounds

T and T'

[00122] When considering the compounds of Formula I, T is optionally substituted alkylene or is absent; and T' is optionally substituted alkylene or is absent. In one embodiment, one of T and T' is absent and the other is optionally substituted alkylene (especially optionally substituted methylene). In another embodiment, both are absent.

R₁

[00123] When considering the compounds of Formula I, in a particular embodiment R₁

is selected from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted heteroaryl, optionally substituted aralkyl, and optionally substituted heteroaralkyl.

In a more particular embodiment R_1 is selected from optionally substituted lower alkyl, optionally substituted aryl, or optionally substituted aralkyl (especially optionally substituted aralkyl).

[00124] In a most particular embodiment R_1 is chosen from ethyl, propyl, methoxyethyl, naphthyl, phenyl, bromophenyl, chlorophenyl, methoxyphenyl, ethoxyphenyl, tolyl, dimethylphenyl, chlorofluorophenyl, methylchlorophenyl, ethylphenyl, phenethyl, benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, hydroxybenzyl, dichlorobenzyl, dimethoxybenzyl, naphthylmethyl, and (ethoxycarbonyl)ethyl. In a more particular embodiment, R_1 is chosen from ethyl, propyl, methoxyethyl, naphthyl, phenethyl, benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, hydroxybenzyl, dichlorobenzyl, dimethoxybenzyl, naphthylmethyl, and (ethoxycarbonyl)ethyl.

[00125] Most particularly, R_1 is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl. Most particularly, R_1 is benzyl.

R_2

[00126] When considering the compounds of Formula I and as will be appreciated by those skilled in the art, the compounds described herein possess a potentially chiral center at the carbon to which R_2 and $R_{2'}$ are attached. The R_2 and $R_{2'}$ groups may be the same or different; if different, the compound is chiral (i.e., has a stereogenic center). When R_2 and $R_{2'}$ are different, in particular embodiments $R_{2'}$ is hydrogen and R_2 is other than hydrogen. The invention contemplates the use of pure enantiomers and mixtures of enantiomers, including racemic mixtures, although the use of a substantially optically pure enantiomer will generally be preferred. The term "substantially optically pure" or "enantiomerically pure" means having at least about 95% of the described enantiomer with no single impurity greater than about 1% and particularly, at least about 97.5% enantiomeric excess. In a particular embodiment, the stereogenic center to which R_2 and $R_{2'}$ are attached is of the R configuration.

[00127] In one embodiment, R_2 is optionally substituted C_1 - C_4 alkyl, and $R_{2'}$ is hydrogen or optionally substituted C_1 - C_4 alkyl. More particularly, $R_{2'}$ is hydrogen and R_2 is optionally substituted C_1 - C_4 alkyl. In a most particular embodiment R_2 is chosen from methyl, ethyl, propyl (particularly, *c*-propyl or *i*-propyl), butyl (particularly, *t*-butyl), methylthioethyl, methylthiomethyl, aminobutyl, (CBZ)aminobutyl, cyclohexylmethyl,

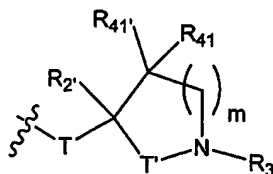
benzyloxymethyl, methylsulfinylethyl, methylsulfinylmethyl, and hydroxymethyl, and $R_{2'}$ is hydrogen. Especially preferred is when $R_{2'}$ is hydrogen and R_2 is ethyl or propyl (particularly, *c*-propyl or *i*-propyl). More particularly, R_2 is *i*-propyl. More preferred is the embodiment wherein the stereogenic center to which R_2 and $R_{2'}$ is attached is of the *R* configuration.

[00128] In another embodiment, both R_2 and $R_{2'}$ are hydrogen.

R_2 taken together with R_7

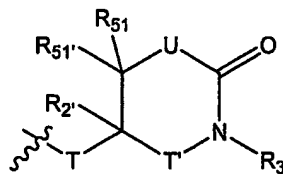
[00129] In another embodiment, R_2 and R_7 taken together form a 5- to 12-membered ring which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring and may optionally be substituted one or more of the following groups: hydroxyl, halogen (particularly chloro and fluoro), optionally substituted C_1 - C_4 alkyl- (particularly methyl-), C_1 - C_4 alkoxy (particularly methoxy), cyano, amino, substituted amino, oxo, or carbamyl.

[00130] In a particular embodiment, R_2 and R_7 taken together form an optionally substituted ring of the formula:



wherein R_{41} and $R_{41'}$ are independently chosen from hydrogen, alkyl, aryl, aralkyl, heteroaryl, substituted alkyl, substituted aryl, substituted aralkyl, and substituted heteroaryl; m is 0, 1, 2, or 3; and T , T' , R_3 , and $R_{2'}$ are as defined above. In a more particular embodiment, R_{41} is hydrogen. In another particular embodiment, both R_{41} and $R_{41'}$ are hydrogen. In another embodiment, R_4 is optionally substituted aralkyl (especially benzyl) or optionally substituted acyl (especially *p*-methyl-benzoyl). See, e.g., PCT application number PCT/US03/30788, filed September 30, 2003, which is incorporated herein by reference for all purposes.

[00131] In another embodiment, R_2 and R_7 taken together form an optionally substituted ring of the formula:



wherein R_3 , R_2' , T , and T' are as defined above; R_{51} and R_{51}' are independently chosen from hydrogen, alkyl, aryl, aralkyl, heteroaryl, substituted alkyl, substituted aryl, substituted aralkyl and substituted heteroaryl; U is a covalent bond, $CR'R''$ or NR''' ; R' and R'' are independently chosen from hydrogen, hydroxy, amino, optionally substituted aryl, optionally substituted alkylamino, optionally substituted alkyl and optionally substituted alkoxy; and R''' is chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, and optionally substituted heteroaralkyl, provided that U and T' are not both covalent bonds.

[00132] In a particular embodiment, R_{51} is hydrogen or optionally substituted lower alkyl; more particularly, R_{51} is hydrogen. In another embodiment, R_{51}' is hydrogen or optionally substituted lower alkyl; more particularly, R_{51}' is hydrogen.

[00133] In one embodiment, R_3 is optionally substituted aryl or optionally substituted aralkyl; more particularly, R_3 is optionally substituted phenyl, benzyl or methyl-benzyl (especially, benzyl or methyl-benzyl).

[00134] In one embodiment, U is $CR'R''$ where R' and/or R'' are hydrogen. In another embodiment, U is NR''' where R''' is hydrogen or optionally substituted alkyl. More particularly, R''' is hydrogen or optionally substituted amino-lower alkyl. See, e.g., USSN 10/626,012 and PCT/US03/22319, each of which is incorporated herein by reference for all purposes.

R_3

[00135] When considering the compounds of Formula I, R_3 is chosen from hydrogen, optionally substituted alkyl-, optionally substituted aryl-, optionally substituted aralkyl-, optionally substituted heteroaryl-, optionally substituted heteroaralkyl-, $-C(O)-R_6$, and $-S(O)_2-R_{6a}$. In one embodiment, R_3 is optionally substituted C_1 - C_{13} alkyl (especially optionally substituted C_1 - C_4 alkyl); optionally substituted aralkyl (especially optionally substituted benzyl or naphthylmethyl-); and optionally substituted heteroaralkyl. More particularly, R_3 is benzyl or benzyl substituted with one or more of the following groups:

carboxy, alkoxycarbonyl cyano, halo, C₁-C₄ alkyl-, C₁-C₄ alkoxy, nitro, methylenedioxy, or trifluoromethyl. In another embodiment, and as described below, R₃ is -C(O)R₆. In yet another embodiment, and as described below R₃ is -SO₂R_{6a}.

R₆ Groups When R₃ is -C(O)R₆

[00136] When considering the compounds of Formula I wherein R₃ is -C(O)R₆, in a particular embodiment R₆ is selected from optionally substituted C₁-C₈ alkyl, optionally substituted aryl-C₁-C₄-alkyl-, optionally substituted heteroaryl-C₁-C₄-alkyl-, optionally substituted heteroaryl, optionally substituted aryl, R₁₁O- and R₁₂-NH-; R₁₁ is chosen from optionally substituted C₁-C₈ alkyl and optionally substituted aryl; and R₁₂ is chosen from hydrogen, optionally substituted C₁-C₈ alkyl and optionally substituted aryl.

[00137] Particular R₆ are selected from optionally substituted C₁-C₈ alkyl, optionally substituted aryl-C₁-C₄-alkyl-, optionally substituted heteroaryl-C₁-C₄-alkyl-, optionally substituted heteroaryl, and optionally substituted aryl. In a more particular embodiment, R₆ is chosen from

phenyl;

phenyl substituted with one or more of the following substituents: halo; C₁-C₄ alkyl; C₁-C₄ alkyl substituted with hydroxy (e.g., hydroxymethyl); C₁-C₄ alkoxy; C₁-C₄ alkyl substituted with C₁-C₄ alkoxy, halo, nitro, formyl, carboxy, cyano, methylenedioxy, ethylenedioxy, acyl (e.g., acetyl), -N-acyl (e.g., N-acetyl) or trifluoromethyl;

benzyl;

phenoxymethyl-;

halophenoxymethyl-;

phenylvinyl-;

heteroaryl-;

heteroaryl- substituted with C₁-C₄ alkyl or C₁-C₄ alkyl substituted with halo (e.g., CF₃);

C₁-C₄ alkyl substituted with C₁-C₄ alkoxy-; and

benzyloxymethyl-.

[00138] In a most particular embodiment, when R₆ is not R₁₂NH- or R₁₁O-, R₆ is chosen from phenyl, halophenyl, dihalophenyl, cyanophenyl, halo(trifluoromethyl)phenyl, hydroxymethyl-phenyl, methoxymethylphenyl, methoxyphenyl, ethoxyphenyl, carboxyphenyl, formylphenyl, ethylphenyl, tolyl, methylenedioxyphenyl,

ethylenedioxyphenyl, methoxychlorophenyl, methylhalophenyl, trifluoromethylphenyl, furanyl, C₁-C₄ alkyl substituted furanyl, trifluoromethylfuranyl, C₁-C₄ alkyl substituted trifluoromethylfuranyl, benzofuranyl, thiophenyl, C₁-C₄ alkyl substituted thiophenyl, benzothiophenyl, benzothiadiazolyl, pyridinyl, indolyl, methylpyridinyl, trifluoromethylpyridinyl, pyrrolyl, quinolinyl, picolinyl, pyrazolyl, C₁-C₄ alkyl substituted pyrazolyl, N-methyl pyrazolyl, C₁-C₄ alkyl substituted N-methyl pyrazolyl, C₁-C₄ alkyl substituted pyrazinyl, C₁-C₄ alkyl substituted isoxazolyl, benzoisoxazolyl, morpholinomethyl, methylthiomethyl, methoxymethyl, N-methyl imidazolyl, and imidazolyl.

Yet more particularly, R₆ is optionally substituted phenyl (especially, tolyl, halophenyl, methylhalophenyl, hydroxymethyl-phenyl, halo(trifluoromethyl)phenyl-, methylenedioxyphenyl, formylphenyl or cyanophenyl).

[00139] In a more particular embodiment, when R₆ is R₁₂NH-, R₁₂ is chosen from hydrogen, C₁-C₄ alkyl; cyclohexyl; phenyl; and phenyl substituted with halo, trifluoromethyl, C₁-C₄ alkyl, C₁-C₄ alkoxy, or C₁-C₄ alkylthio-.

[00140] In a most particular embodiment, when R₆ is R₁₂NH-, R₁₂ is hydrogen, isopropyl, butyl, cyclohexyl, phenyl, bromophenyl, dichlorophenyl, methoxyphenyl, ethylphenyl, tolyl, trifluoromethylphenyl or methylthio-phenyl.

[00141] In an embodiment, wherein R₆ is R₁₁O-, R₁₁ is chosen from optionally substituted C₁-C₈ alkyl and optionally substituted aryl.

R_{6a} Groups when R₃ is -SO₂R_{6a}

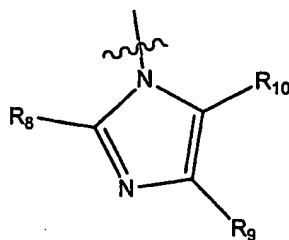
[00142] In one embodiment, when R₃ is -SO₂R_{6a}, R_{6a} is chosen from C₁-C₁₃ alkyl; phenyl; naphthyl; phenyl substituted with halo, lower alkyl, lower alkoxy, cyano, nitro, methylenedioxy, or trifluoromethyl; biphenyl and heteroaryl. More particularly, R_{6a} is chosen from phenyl substituted with halo, lower alkyl, lower alkoxy, cyano, nitro, methylenedioxy, or trifluoromethyl and naphthyl.

R₃ taken together with R₇

[00143] When considering the compounds of Formula I, in one embodiment, R₃ taken together with R₇, and the nitrogen to which they are bound, form an optionally substituted 5- to 12-membered nitrogen-containing heterocycle, which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring.

[00144] In a particular embodiment, R₃ taken together with R₇ and the nitrogen to

which they are bound, forms an optionally substituted imidazolyl ring of the formula:



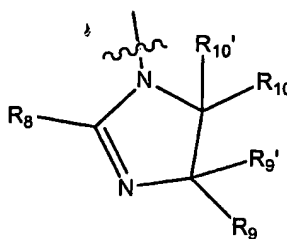
wherein

R_8 is chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaralkyl, optionally substituted aralkoxy, optionally substituted heteroaralkoxy, and optionally substituted heteroaryl; and

R_9 and R_{10} are independently hydrogen, optionally substituted alkyl, optionally substituted aryl, or optionally substituted aralkyl.

[00145] More particularly, when R_3 taken together with R_7 and the nitrogen to which they are bound, forms an optionally substituted imidazolyl ring, R_8 is aryl (especially phenyl), substituted aryl (especially lower alkyl-, lower alkoxy-, and/or halo-substituted phenyl), aralkyl (especially benzyl and phenylvinyl), heteroaryl, substituted heteroaryl, heteroaralkyl, aralkoxy (especially phenoxy lower alkyl), heteroaralkoxy, substituted aralkyl (especially substituted benzyl and substituted styrenyl), substituted heteroaralkyl, substituted aralkoxy (especially substituted phenoxy lower alkyl), or substituted heteroaralkoxy. See, e.g., USSN 10/435,069 and PCT/US03/14787, each of which is incorporated herein by reference.

[00146] In another particular embodiment, R_3 taken together with R_7 forms an optionally substituted imidazolinyl ring of the formula:



wherein,

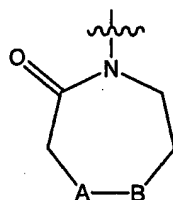
R_8 is chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, optionally substituted heteroaralkyl, optionally substituted aralkoxy, or optionally substituted heteroaralkoxy; and

R_{10} , $R_{10'}$, R_9 , and $R_{9'}$ are independently chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, and optionally substituted aralkyl.

[00147] When R_3 taken together with R_7 forms an optionally substituted imidazoliny ring, in a particular embodiment, R_8 is aryl (especially phenyl), substituted aryl (especially lower alkyl-, lower alkoxy-, and/or halo-substituted phenyl), aralkyl (especially benzyl and phenylvinyl), heteroaryl, substituted heteroaryl, heteroaralkyl, aralkoxy (especially phenoxy lower alkyl), heteroaralkoxy, substituted aralkyl (especially substituted benzyl and substituted styrenyl), substituted heteroaralkyl, substituted aralkoxy (especially substituted phenoxy lower alkyl), or substituted heteroaralkoxy.

[00148] When R_3 taken together with R_7 forms an optionally substituted imidazoliny ring, more particularly, R_9 is hydrogen or optionally substituted lower alkyl, and $R_{9'}$ is hydrogen or optionally substituted lower alkyl.

[00149] In another embodiment, R_3 taken together with R_7 forms an optionally substituted diazepinone ring of the formula:

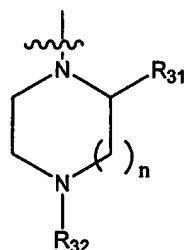


wherein A and B are each independently chosen from $C(R_{20})(R_{21})$, $N(R_{22})$, O or S, wherein R_{20} and R_{21} are each independently selected from H, optionally substituted alkyl optionally substituted aryl and optionally substituted heteroaryl; and R_{22} is H, optionally substituted alkyl, optionally substituted aralkyl, optionally substituted heteroaralkyl, optionally substituted alkylcarbonyl, optionally substituted arylcarbonyl, optionally substituted heteroarylcarbonyl, optionally substituted aralkylcarbonyl, optionally substituted heteroaralkylcarbonyl, optionally substituted alkoxy carbonyl, optionally substituted aryloxy carbonyl, optionally substituted heteroaryloxy carbonyl, optionally substituted

aralkyloxycarbonyl, optionally substituted heteroaralkyloxycarbonyl. In a more particular embodiment, the diazepinone ring is further substituted with one or more of the following groups: optionally substituted alkyl, optionally substituted aryl, optionally substituted aralkyl, optionally substituted heteroaryl, and optionally substituted heteroaralkyl.

[00150] In yet another embodiment of the compounds of Formula I, one of A or B is $C(R_{20})(R_{21})$, wherein R_{20} and R_{21} are each independently selected from H or C_1 - C_4 alkyl, and the other of A or B is $N(R_{22})$, where R_{22} is H, C_1 - C_4 alkyl, optionally substituted aralkyl, optionally substituted heteroaralkyl, C_1 - C_6 alkylcarbonyl, optionally substituted arylcarbonyl, optionally substituted heteroarylcarbonyl, optionally substituted aralkylcarbonyl, optionally substituted heteroaralkylcarbonyl, C_1 - C_6 alkoxy, optionally substituted aryloxycarbonyl, optionally substituted heteroaryloxycarbonyl, optionally substituted aralkyloxycarbonyl, optionally substituted heteroaralkyloxycarbonyl, where the optionally substituted aryl or heteroaryl groups or moieties are unsubstituted or substituted with one or more substituents selected from C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_1 - C_4 alkoxy, C_1 - C_4 haloalkoxy, amino, C_1 - C_4 alkylamino, di- C_1 - C_4 alkylamino, carboxy, C_1 - C_4 alkylcarbonyloxy, C_1 - C_4 alkoxy, carboxamido, C_1 - C_4 alkylcarboxamido, aminocarbonyl, C_1 - C_4 alkylaminocarbonyl, di- C_1 - C_4 alkylaminocarbonyl, cyano, C_1 - C_4 alkylcarbonyl, halogen, hydroxyl, mercapto and nitro. In another embodiment, A is $C(R_{20})(R_{21})$, wherein R_{20} and R_{21} are each H or C_1 - C_4 alkyl, and B is $N(R_{22})$, where R_{22} is H, C_1 - C_4 alkyl, aralkyl, heteroaralkyl, C_1 - C_6 alkylcarbonyl, arylcarbonyl, heteroarylcarbonyl. In specific embodiments of the compounds of Formula I, A is CH_2 , and B is $N(R_{22})$, where R_{22} is H, methyl, benzyl or acetyl ($-C(O) \text{methyl}$). See, e.g., USSN 60/435,001, which is incorporated herein by reference for all purposes.

[00151] In another embodiment, R_3 taken together with R_7 forms an optionally substituted piperazine- or diazepam of the formula:



R₃₁ and R₃₂ are independently chosen from hydrogen, optionally substituted alkyl, optionally substituted aryl, optionally substituted heteroaryl, optionally substituted aralkyl, and optionally substituted heteroaralkyl; and n is 1 or 2. More particularly, R₃₁ is aryl (especially phenyl), substituted aryl (especially lower alkyl-, lower alkoxy-, and/or halo-substituted phenyl), aralkyl (especially benzyl and phenylvinyl), heteroaralkyl, substituted aralkyl (especially substituted benzyl and substituted phenylvinyl), or substituted heteroaralkyl; R₃₂ is hydrogen; and n is 1. See, e.g., USSN 10/644,244 and PCT/US03/26093, each of which is incorporated herein by reference.

R₇

[00152] When considering compounds of Formula I, in a particular embodiment, R₇ is chosen from hydrogen, optionally substituted C₁-C₁₃ alkyl, optionally substituted aryl, optionally substituted aryl-C₁-C₄-alkyl-, optionally substituted heterocyclyl, and optionally substituted heteroaryl-C₁-C₄-alkyl- (especially hydrogen or optionally substituted C₁-C₁₃ alkyl).

[00153] More particularly, R₇ is chosen from hydrogen, C₁-C₄ alkyl; cyclohexyl; phenyl substituted with hydroxyl, C₁-C₄ alkoxy or C₁-C₄ alkyl; benzyl; and R₁₆-alkylene-, wherein R₁₆ is hydroxyl, carboxy, (C₁-C₄ alkoxy)carbonyl-, di(C₁-C₄ alkyl)amino-, (C₁-C₄ alkyl)amino-, amino, (C₁-C₄ alkoxy)carbonylamino-, C₁-C₄ alkoxy-, or optionally substituted N-heterocyclyl- (particularly azetidiny, morpholinyl, pyridinyl, indolyl, furanyl, pyrrolidinyl, piperidinyl or imidazolyl, each of which may be optionally substituted).

[00154] In a particular embodiment, R₇ is chosen from hydrogen, methyl, ethyl, propyl, butyl, cyclohexyl, carboxyethyl, carboxymethyl, methoxyethyl, hydroxyethyl, hydroxypropyl, dimethylaminoethyl, dimethylaminopropyl, diethylaminoethyl, diethylaminopropyl, aminopropyl, methylaminopropyl, 2,2-dimethyl-3-(dimethylamino)propyl, aminoethyl, aminobutyl, aminopentyl, aminohexyl, isopropylaminopropyl, diisopropylaminoethyl, 1-methyl-4-(diethylamino)butyl, (t-Boc)aminopropyl, hydroxyphenyl, benzyl, methoxyphenyl, methylmethoxyphenyl, dimethylphenyl, tolyl, ethylphenyl, (oxopyrrolidinyl)propyl, (methoxycarbonyl)ethyl, benzylpiperidinyl, pyridinylethyl, pyridinylmethyl, morpholinylethyl, morpholinylpropyl, piperidinyl, azetidinylmethyl, azetidinylethyl, azetidinypropyl, pyrrolidinylethyl, pyrrolidinylpropyl, piperidinylmethyl, piperidinylethyl, imidazolylpropyl, imidazolylethyl, (ethylpyrrolidinyl)methyl, (methylpyrrolidinyl)ethyl,

(methylpiperidinyl)propyl, (methylpiperazinyl)propyl, furanylmethyl and indolyethyl.

[00155] In another embodiment, R_7 is R_{16} -alkylene-, wherein R_{16} is amino, C_1 - C_4 alkylamino-, di(C_1 - C_4 alkyl)amino-, C_1 - C_4 alkoxy-, hydroxyl, or N-heterocyclyl.

Particularly R_{16} is amino. In a particular embodiment, the alkylene moiety of R_{16} -alkylene- has from 1 to 6 carbon atoms.

[00156] More particularly, R_7 is aminoethyl, aminopropyl, aminobutyl, aminopentyl, aminohexyl, methylaminoethyl, methylaminopropyl, methylaminobutyl, methylaminopentyl, methylaminohexyl, dimethylaminoethyl, dimethylaminopropyl, dimethylaminobutyl, dimethylaminopentyl, dimethylaminohexyl, ethylaminoethyl, ethylaminopropyl, ethylaminobutyl, ethylaminopentyl, ethylaminohexyl, diethylaminoethyl, diethylaminopropyl, diethylaminobutyl, diethylaminopentyl, or diethylaminohexyl, most particularly aminopropyl.

R_4 and R_5

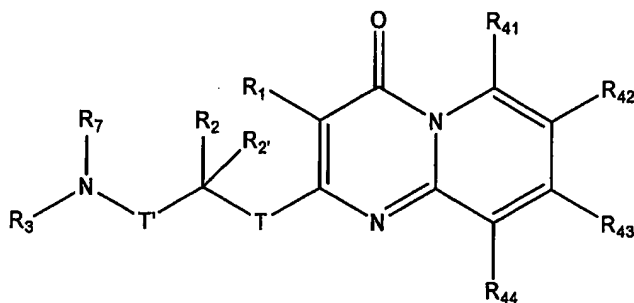
[00157] When considering the compounds of Formula I, in one embodiment, R_4 is chosen from hydrogen, hydroxyl, optionally substituted lower alkyl (particularly methyl), lower alkoxy (particularly methoxy) and cyano. More preferably, R_4 is hydrogen, optionally substituted alkyl (particularly, methyl), optionally substituted aryl (particularly, phenyl), alkoxy (particularly, methoxy), cyano, substituted amino, carbamyl, aryloxy (particularly, phenoxy), heteroaryloxy (particularly, pyridinyloxy), heteroaryl (particularly, 2-oxo-2H-pyridinyl), optionally substituted N-heterocyclyl (particularly, morpholinyl or piperazinyl), or trifluoromethyl.

[00158] When considering the compounds of Formula I, in one embodiment, R_5 is chosen from hydrogen, optionally substituted lower alkyl (particularly methyl), optionally substituted aryl, and optionally substituted aralkyl (particularly benzyl).

[00159] In another embodiment, R_4 and R_5 taken together form an optionally substituted 5 to 7-membered nitrogen-containing heterocycle which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring. The ring may be aromatic or non-aromatic. In a more particular embodiment, the heterocycle is a pyridinyl, pyridazinyl, pyrimidinyl, pyrazinyl, piperidinyl, piperazinyl, hexahydropyrimidinyl, piperazinyl, morpholinyl, pyrrolyl, pyrazolyl, imidazolyl, dihydroisoxazolyl, or dihydrooxazolyl ring which is optionally substituted with one or more of the following groups (and especially one of the following groups): hydroxyl, halogen

(particularly chloro and fluoro), optionally substituted C₁-C₄ alkyl- (particularly methyl-), C₁-C₄ alkoxy (particularly methoxy), cyano, amino, substituted amino, oxo, or carbamyl.

[00160] In a more particular embodiment, R₄ and R₅ taken together form an optionally substituted pyridinyl ring, i.e., a compound of Formula IA.:



Formula IA

[00161] wherein R₁, R₂, R_{2'}, R₃, R₇, T and T' are as described above and R₄₁, R₄₂, R₄₃, and R₄₄ are independently chosen from hydrogen, optionally substituted alkyl, optionally substituted alkoxy, halogen, hydroxyl, nitro, cyano, dialkylamino, alkylsulfonyl, alkylsulfonamido, alkylthio, carboxyalkyl, carboxamido, aminocarbonyl, optionally substituted aryl and optionally substituted heteroaryl. More particularly, R₄₁, R₄₂, R₄₃, and R₄₄ are independently chosen from hydrogen, hydroxyl, halo (particularly chloro and fluoro), lower alkyl (particularly methyl), lower alkoxy (particularly methoxy) and cyano. More preferably, R₄₁, R₄₂, R₄₃, and R₄₄ are methoxy, hydrogen or halo. Further preferred for each of the specific substituents: R₄₁ is hydrogen or halo; R₄₂ is hydrogen, alkyl (particularly, methyl) or halo; R₄₃ is hydrogen, halo, alkyl (particularly, methyl), alkoxy (particularly, methoxy), cyano, or trifluoromethyl; and R₄₄ is hydrogen or halo. Still further preferred are the compounds where only one of R₄₁, R₄₂, R₄₃, and R₄₄ is not hydrogen, especially R₄₃.

Salt Forms

[00162] Compounds of the invention will generally be capable of forming acid addition salts (i.e., will comprise a site which reacts with a pharmaceutically acceptable acid to form an acid addition salt.) The present invention includes pharmaceutically acceptable acid addition salts of the compounds of Formula I. Acid addition salts of the present compounds

are prepared in a standard manner in a suitable solvent from the parent compound and an excess of an acid, such as hydrochloric, hydrobromic, sulfuric, phosphoric, acetic, maleic, succinic or methanesulfonic.

[00163] The salts and/or solvates of the compounds of the formula I which are not pharmaceutically acceptable may be useful as intermediates in the preparation of pharmaceutically acceptable salts and/or solvates of compounds of formula I or the compounds of the formula I themselves, and as such form another aspect of the present invention.

Particular Subgenus of Compounds of Formula I

[00164] When considering the compounds of Formula I, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is -C(O)R₆;

R₆ is optionally substituted phenyl;

R₇ is R₁₆-alkylene-;

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl;

R₄ is chosen from hydrogen, hydroxyl, lower alkyl (particularly methyl), lower alkoxy (particularly methoxy) and cyano; and

R₅ is chosen from hydrogen, lower alkyl (particularly methyl), and aralkyl (particularly benzyl).

[00165] When considering the compounds of Formula I, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is -C(O)R₆;

R₆ is R₁₂NH-;

R₁₂ is chosen from hydrogen, C₁-C₄ alkyl; cyclohexyl; and optionally substituted phenyl;

R₇ is R₁₆-alkylene-,

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl;

R₄ is chosen from hydrogen, hydroxyl, lower alkyl (particularly methyl), lower alkoxy (particularly methoxy) and cyano; and

R₅ is chosen from hydrogen, lower alkyl (particularly methyl), and aralkyl (particularly benzyl).

[00166] When considering the compounds of Formula I, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is -C(O)R₆;

R₆ is R₁₁O-;

R₁₁ is chosen from optionally substituted C₁-C₈ alkyl and optionally substituted aryl;

R₇ is R₁₆-alkylene-;

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl;

R₄ is chosen from hydrogen, hydroxyl, lower alkyl (particularly methyl), lower alkoxy (particularly methoxy) and cyano; and

R₅ is chosen from hydrogen, lower alkyl (particularly methyl), and aralkyl (particularly benzyl).

[00167] When considering the compounds of Formula I, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is -SO₂R_{6a};

R_{6a} is chosen from phenyl substituted with halo, lower alkyl, lower alkoxy, cyano, nitro, methylenedioxy, or trifluoromethyl and naphthyl;

R₇ is R₁₆-alkylene-,

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl;

R₄ is chosen from hydrogen, hydroxyl, lower alkyl (particularly methyl), lower alkoxy (particularly methoxy) and cyano; and

R₅ is chosen from hydrogen, lower alkyl (particularly methyl), and aralkyl (particularly benzyl).

[00168] When considering the compounds of Formula I, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is optionally substituted C₁-C₁₃ alkyl (especially optionally substituted C₁-C₄ alkyl); optionally substituted aralkyl (especially optionally substituted benzyl or naphthylmethyl-); and optionally substituted heteroaralkyl;

R₇ is R₁₆-alkylene-,

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl;

R₄ is chosen from hydrogen, hydroxyl, lower alkyl (particularly methyl), lower alkoxy (particularly methoxy) and cyano; and

R₅ is chosen from hydrogen, lower alkyl (particularly methyl), and aralkyl (particularly benzyl).

[00169] When considering the compounds of Formula I, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ taken together with R₇, and the nitrogen to which they are bound, form an optionally substituted 5- to 12-membered nitrogen-containing heterocycle, which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring.;

R₄ is chosen from hydrogen, hydroxyl, lower alkyl (particularly methyl), lower alkoxy (particularly methoxy) and cyano; and

R₅ is chosen from hydrogen, lower alkyl (particularly methyl), and aralkyl (particularly benzyl).

[00170] When considering the compounds of Formula I or Ia, in a particular

embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is -C(O)R₆;

R₆ is optionally substituted phenyl;

R₇ is R₁₆-alkylene-,

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl; and

R₄ and R₅ taken together form an optionally substituted 5 to 7-membered nitrogen-containing heterocycle which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring.

[00171] When considering the compounds of Formula I or Ia, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is -C(O)R₆;

R₆ is R₁₂NH-;

R₁₂ is chosen from hydrogen, C₁-C₄ alkyl; cyclohexyl; and optionally substituted phenyl;

R₇ is R₁₆-alkylene-,

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl; and

R₄ and R₅ taken together form an optionally substituted 5 to 7-membered nitrogen-containing heterocycle which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring.

[00172] When considering the compounds of Formula I or Ia, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂' is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is -C(O)R₆;

R₆ is R₁₁O-;

R₁₁ is chosen from optionally substituted C₁-C₈ alkyl and optionally substituted aryl;

R₇ is R₁₆-alkylene-,

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl; and

R₄ and R₅ taken together form an optionally substituted 5 to 7-membered nitrogen-containing heterocycle which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring.

[00173] When considering the compounds of Formula I or Ia, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂' is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is -SO₂R_{6a};

R_{6a} is chosen from phenyl substituted with halo, lower alkyl, lower alkoxy, cyano, nitro, methylenedioxy, or trifluoromethyl and naphthyl;

R₇ is R₁₆-alkylene-,

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl; and

R₄ and R₅ taken together form an optionally substituted 5 to 7-membered nitrogen-containing heterocycle which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring.

[00174] When considering the compounds of Formula I or Ia, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂' is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ is optionally substituted C₁-C₁₃ alkyl (especially optionally substituted C₁-C₄

alkyl); optionally substituted aralkyl (especially optionally substituted benzyl or naphthylmethyl-); and optionally substituted heteroaralkyl;

R₇ is R₁₆-alkylene-,

R₁₆ is amino, C₁-C₄ alkylamino-, di(C₁-C₄ alkyl)amino-, C₁-C₄ alkoxy-, hydroxyl, or N-heterocyclyl; and

R₄ and R₅ taken together form an optionally substituted 5 to 7-membered nitrogen-containing heterocycle which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring.

[00175] When considering the compounds of Formula I or Ia, in a particular embodiment,

R₁ is benzyl, chlorobenzyl, methylbenzyl, methoxybenzyl, cyanobenzyl, or hydroxybenzyl;

R₂ is hydrogen;

R₂ is optionally substituted C₁-C₄ alkyl;

R₃ taken together with R₇, and the nitrogen to which they are bound, form an optionally substituted 5- to 12-membered nitrogen-containing heterocycle, which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring; and

R₄ and R₅ taken together form an optionally substituted 5 to 7-membered nitrogen-containing heterocycle which optionally incorporates from one to two additional heteroatoms, selected from N, O, and S in the heterocycle ring.

[00176] Particular compounds of the invention are:

N-(3-Amino-propyl)-N-[1-(5-benzyl-2-methyl-6-oxo-1,6-dihydro-pyrimidin-4-yl)-2-methyl-propyl]-4-methyl-benzamide;

N-(3-Amino-propyl)-N-[1-(5-benzyl-1,2-dimethyl-6-oxo-1,6-dihydro-pyrimidin-4-yl)-2-methyl-propyl]-4-methyl-benzamide;

N-(3-Amino-propyl)-N-[1-(3-benzyl-4-oxo-4H-pyrido[1,2-a]pyrimidin-2-yl)-2-methyl-propyl]-4-methyl-benzamide; and

N-(3-Amino-propyl)-N-[1-(3-benzyl-8-chloro-4-oxo-4H-pyrido[1,2-a]pyrimidin-2-yl)-2-methyl-propyl]-4-methyl-benzamide.

Utility, Testing and Administration

General Utility

[00177] Once made, the compounds of the invention find use in a variety of applications involving alteration of mitosis. As will be appreciated by those skilled in the art, mitosis may be altered in a variety of ways; that is, one can affect mitosis either by increasing or decreasing the activity of a component in the mitotic pathway. Stated differently, mitosis may be affected (e.g., disrupted) by disturbing equilibrium, either by inhibiting or activating certain components. Similar approaches may be used to alter meiosis.

[00178] In a particular embodiment, the compounds of the invention are used to inhibit mitotic spindle formation, thus causing prolonged cell cycle arrest in mitosis. By "inhibit" in this context is meant decreasing or interfering with mitotic spindle formation or causing mitotic spindle dysfunction. By "mitotic spindle formation" herein is meant organization of microtubules into bipolar structures by mitotic kinesins. By "mitotic spindle dysfunction" herein is meant mitotic arrest and monopolar spindle formation.

[00179] The compounds of the invention are useful to bind to, and/or inhibit the activity of, a mitotic kinesin, KSP. In one embodiment, the KSP is human KSP, although the compounds may be used to bind to or inhibit the activity of KSP kinesins from other organisms. In this context, "inhibit" means either increasing or decreasing spindle pole separation, causing malformation, i.e., splaying, of mitotic spindle poles, or otherwise causing morphological perturbation of the mitotic spindle. Also included within the definition of KSP for these purposes are variants and/or fragments of KSP. See U.S. Patent 6,437,115, hereby incorporated by reference in its entirety. The compounds of the invention have been shown to have specificity for KSP. However, the present invention includes the use of the compounds to bind to or modulate other mitotic kinesins.

[00180] The compounds of the invention are used to treat cellular proliferation diseases. Such disease states which can be treated by the compounds, compositions and methods provided herein include, but are not limited to, cancer (further discussed below), autoimmune disease, fungal disorders, arthritis, graft rejection, inflammatory bowel disease, cellular proliferation induced after medical procedures, including, but not limited to, surgery, angioplasty, and the like. Treatment includes inhibiting cellular proliferation. It is appreciated that in some cases the cells may not be in an abnormal state and still require treatment. Thus, in one embodiment, the invention herein includes application to cells or individuals afflicted or subject to impending affliction with any one of these disorders or states.

[00181] The compounds, pharmaceutical formulations and methods provided herein

are particularly deemed useful for the treatment of cancer including solid tumors such as skin, breast, brain, cervical carcinomas, testicular carcinomas, etc. More particularly, cancers that can be treated include, but are not limited to:

- Cardiac: sarcoma (angiosarcoma, fibrosarcoma, rhabdomyosarcoma, liposarcoma), myxoma, rhabdomyoma, fibroma, lipoma and teratoma;
- Lung: bronchogenic carcinoma (squamous cell, undifferentiated small cell, undifferentiated large cell, adenocarcinoma), alveolar (bronchiolar) carcinoma, bronchial adenoma, sarcoma, lymphoma, chondromatous hamartoma, mesothelioma;
- Gastrointestinal: esophagus (squamous cell carcinoma, adenocarcinoma, leiomyosarcoma, lymphoma), stomach (carcinoma, lymphoma, leiomyosarcoma), pancreas (ductal adenocarcinoma, insulinoma, glucagonoma, gastrinoma, carcinoid tumors, vipoma), small bowel (adenocarcinoma, lymphoma, carcinoid tumors, Kaposi's sarcoma, leiomyoma, hemangioma, lipoma, neurofibroma, fibroma), large bowel (adenocarcinoma, tubular adenoma, villous adenoma, hamartoma, leiomyoma);
- Genitourinary tract: kidney (adenocarcinoma, Wilm's tumor [nephroblastoma], lymphoma, leukemia), bladder and urethra (squamous cell carcinoma, transitional cell carcinoma, adenocarcinoma), prostate (adenocarcinoma, sarcoma), testis (seminoma, teratoma, embryonal carcinoma, teratocarcinoma, choriocarcinoma, sarcoma, interstitial cell carcinoma, fibroma, fibroadenoma, adenomatoid tumors, lipoma);
- Liver: hepatoma (hepatocellular carcinoma), cholangiocarcinoma, hepatoblastoma, angiosarcoma, hepatocellular adenoma, hemangioma;
- Bone: osteogenic sarcoma (osteosarcoma), fibrosarcoma, malignant fibrous histiocytoma, chondrosarcoma, Ewing's sarcoma, malignant lymphoma (reticulum cell sarcoma), multiple myeloma, malignant giant cell tumor chordoma, osteochondroma (osteochondrogenous exostoses), benign chondroma, chondroblastoma, chondromyxofibroma, osteoid osteoma and giant cell tumors;
- Nervous system: skull (osteoma, hemangioma, granuloma, xanthoma, osteitis deformans), meninges (meningioma, meningiosarcoma, gliomatosis), brain (astrocytoma, medulloblastoma, glioma, ependymoma, germinoma [pinealoma], glioblastoma multiform, oligodendroglioma, schwannoma, retinoblastoma, congenital tumors), spinal cord neurofibroma, meningioma, glioma, sarcoma);
- Gynecological: uterus (endometrial carcinoma), cervix (cervical carcinoma, pre-tumor

cervical dysplasia), ovaries (ovarian carcinoma [serous cystadenocarcinoma, mucinous cystadenocarcinoma, unclassified carcinoma], granulosa-thecal cell tumors, Sertoli-Leydig cell tumors, dysgerminoma, malignant teratoma), vulva (squamous cell carcinoma, intraepithelial carcinoma, adenocarcinoma, fibrosarcoma, melanoma), vagina (clear cell carcinoma, squamous cell carcinoma, botryoid sarcoma (embryonal rhabdomyosarcoma), fallopian tubes (carcinoma);

- Hematologic: blood (myeloid leukemia [acute and chronic], acute lymphoblastic leukemia, chronic lymphocytic leukemia, myeloproliferative diseases, multiple myeloma, myelodysplastic syndrome), Hodgkin's disease, non-Hodgkin's lymphoma [malignant lymphoma];
- Skin: malignant melanoma, basal cell carcinoma, squamous cell carcinoma, Kaposi's sarcoma, moles dysplastic nevi, lipoma, angioma, dermatofibroma, keloids, psoriasis; and
- Adrenal glands: neuroblastoma.

As used herein, treatment of cancer includes treatment of cancerous cells, including cells afflicted by any one of the above-identified conditions. Thus, the term "cancerous cell" as provided herein, includes a cell afflicted by any one of the above identified conditions.

[00182] Another useful aspect of the invention is a kit having a compound, salt or solvate of Formula I and a package insert or other labeling including directions treating a cellular proliferative disease by administering an effective amount of the compound, salt or solvate. The compound, salt or solvate of Formula I in the kits of the invention is particularly provided as one or more doses for a course of treatment for a cellular proliferative disease, each dose being a pharmaceutical formulation including a pharmaceutical excipient and a compound, salt or solvate of Formula I.

Testing

[00183] For assay of KSP-modulating activity, generally either KSP or a compound according to the invention is non-diffusably bound to an insoluble support having isolated sample receiving areas (e.g., a microtiter plate, an array, etc.). The insoluble support may be made of any composition to which the sample can be bound, is readily separated from soluble material, and is otherwise compatible with the overall method of screening. The surface of such supports may be solid or porous and of any convenient shape. Examples of suitable insoluble supports include microtiter plates, arrays, membranes and beads. These are typically made of glass, plastic (e.g., polystyrene), polysaccharides, nylon or nitrocellulose,

Teflon™, etc. Microtiter plates and arrays are especially convenient because a large number of assays can be carried out simultaneously, using small amounts of reagents and samples. The particular manner of binding of the sample is not crucial so long as it is compatible with the reagents and overall methods of the invention, maintains the activity of the sample and is nondiffusible. Particular methods of binding include the use of antibodies (which do not sterically block either the ligand binding site or activation sequence when the protein is bound to the support), direct binding to "sticky" or ionic supports, chemical crosslinking, the synthesis of the protein or agent on the surface, etc. Following binding of the sample, excess unbound material is removed by washing. The sample receiving areas may then be blocked through incubation with bovine serum albumin (BSA), casein or other innocuous protein or other moiety.

[00184] The compounds of the invention may be used on their own to inhibit the activity of a mitotic kinesin, particularly KSP. In one embodiment, a compound of the invention is combined with KSP and the activity of KSP is assayed. Kinesin (including KSP) activity is known in the art and includes one or more kinesin activities. Kinesin activities include the ability to affect ATP hydrolysis; microtubule binding; gliding and polymerization/depolymerization (effects on microtubule dynamics); binding to other proteins of the spindle; binding to proteins involved in cell-cycle control; serving as a substrate to other enzymes, such as kinases or proteases; and specific kinesin cellular activities such as spindle pole separation.

[00185] Methods of performing motility assays are well known to those of skill in the art. (See e.g., Hall, et al. (1996), *Biophys. J.*, 71: 3467-3476, Turner et al., 1996, *Anal. Biochem.* 242 (1):20-5; Gittes et al., 1996, *Biophys. J.* 70(1): 418-29; Shirakawa et al., 1995, *J. Exp. Biol.* 198: 1809-15; Winkelmann et al., 1995, *Biophys. J.* 68: 2444-53; Winkelmann et al., 1995, *Biophys. J.* 68: 72S.)

[00186] Methods known in the art for determining ATPase hydrolysis activity also can be used. Suitably, solution based assays are utilized. U.S. Patent 6,410,254, hereby incorporated by reference in its entirety, describes such assays. Alternatively, conventional methods are used. For example, P_i release from kinesin can be quantified. In one embodiment, the ATPase hydrolysis activity assay utilizes 0.3 M PCA (perchloric acid) and malachite green reagent (8.27 mM sodium molybdate II, 0.33 mM malachite green oxalate, and 0.8 mM Triton X-100). To perform the assay, 10 μ L of the reaction mixture is quenched in 90 μ L of cold 0.3 M PCA. Phosphate standards are used so data can be converted to mM

inorganic phosphate released. When all reactions and standards have been quenched in PCA, 100 μ L of malachite green reagent is added to the relevant wells in e.g., a microtiter plate. The mixture is developed for 10-15 minutes and the plate is read at an absorbance of 650 nm. If phosphate standards were used, absorbance readings can be converted to mM P_i and plotted over time. Additionally, ATPase assays known in the art include the luciferase assay.

[00187] ATPase activity of kinesin motor domains also can be used to monitor the effects of agents and are well known to those skilled in the art. In one embodiment ATPase assays of kinesin are performed in the absence of microtubules. In another embodiment, the ATPase assays are performed in the presence of microtubules. Different types of agents can be detected in the above assays. In a one embodiment, the effect of an agent is independent of the concentration of microtubules and ATP. In another embodiment, the effect of the agents on kinesin ATPase can be decreased by increasing the concentrations of ATP, microtubules or both. In yet another embodiment, the effect of the agent is increased by increasing concentrations of ATP, microtubules or both.

[00188] Compounds that inhibit the biochemical activity of KSP in vitro may then be screened in vivo. In vivo screening methods include assays of cell cycle distribution, cell viability, or the presence, morphology, activity, distribution, or number of mitotic spindles. Methods for monitoring cell cycle distribution of a cell population, for example, by flow cytometry, are well known to those skilled in the art, as are methods for determining cell viability. See for example, U.S. Patent 6,437,115, hereby incorporated by reference in its entirety. Microscopic methods for monitoring spindle formation and malformation are well known to those of skill in the art (see, e.g., Whitehead and Rattner (1998), *J. Cell Sci.* 111:2551-61; Galgio et al, (1996) *J. Cell Biol.*, 135:399-414), each incorporated herein by reference in its entirety.

[00189] The compounds of the invention inhibit the KSP kinesin. One measure of inhibition is IC_{50} , defined as the concentration of the compound at which the activity of KSP is decreased by fifty percent relative to a control. Preferred compounds have IC_{50} 's of less than about 1 mM, with preferred embodiments having IC_{50} 's of less than about 100 μ M, with more preferred embodiments having IC_{50} 's of less than about 10 μ M, with particularly preferred embodiments having IC_{50} 's of less than about 1 μ M, and especially preferred embodiments having IC_{50} 's of less than about 100 nM, and with the most preferred embodiments having IC_{50} 's of less than about 10 nM. Measurement of IC_{50} is done using an

ATPase assay such as described herein.

[00190] Another measure of inhibition is K_i . For compounds with IC_{50} 's less than 1 μM , the K_i or K_d is defined as the dissociation rate constant for the interaction of the compounds described herein with KSP. Preferred compounds have K_i 's of less than about 100 μM , with preferred embodiments having K_i 's of less than about 10 μM , and particularly preferred embodiments having K_i 's of less than about 1 μM and especially preferred embodiments having K_i 's of less than about 100 nM, and with the most preferred embodiments having K_i 's of less than about 10 nM.

[00191] The K_i for a compound is determined from the IC_{50} based on three assumptions and the Michaelis-Menten equation. First, only one compound molecule binds to the enzyme and there is no cooperativity. Second, the concentrations of active enzyme and the compound tested are known (i.e., there are no significant amounts of impurities or inactive forms in the preparations). Third, the enzymatic rate of the enzyme-inhibitor complex is zero. The rate (i.e., compound concentration) data are fitted to the equation:

$$V = V_{\max} E_0 \left[1 - \frac{(E_0 + I_0 + K_d) - \sqrt{(E_0 + I_0 + K_d)^2 - 4 E_0 I_0}}{2 E_0} \right]$$

where V is the observed rate, V_{\max} is the rate of the free enzyme, I_0 is the inhibitor concentration, E_0 is the enzyme concentration, and K_d is the dissociation constant of the enzyme-inhibitor complex.

[00192] Another measure of inhibition is GI_{50} , defined as the concentration of the compound that results in a decrease in the rate of cell growth by fifty percent. Preferred compounds have GI_{50} 's of less than about 1 mM; those having a GI_{50} of less than about 20 μM are more preferred; those having a GI_{50} of less than about 10 μM more so; those having a GI_{50} of less than about 1 μM more so; those having a GI_{50} of less than about 100 nM more so; and those having a GI_{50} of less than about 10 nM even more so. Measurement of GI_{50} is done using a cell proliferation assay such as described herein. Compounds of this class were found to inhibit cell proliferation.

[00193] In vitro potency of small molecule inhibitors is determined, for example, by assaying human ovarian cancer cells (SKOV3) for viability following a 72-hour exposure to a 9-point dilution series of compound. Cell viability is determined by measuring the absorbance of formazon, a product formed by the bioreduction of MTS/PMS, a commercially

available reagent. Each point on the dose-response curve is calculated as a percent of untreated control cells at 72 hours minus background absorption (complete cell kill).

[00194] Anti-proliferative compounds that have been successfully applied in the clinic to treatment of cancer (cancer chemotherapeutics) have GI_{50} 's that vary greatly. For example, in A549 cells, paclitaxel GI_{50} is 4 nM, doxorubicin is 63 nM, 5-fluorouracil is 1 μ M, and hydroxyurea is 500 μ M (data provided by National Cancer Institute, Developmental Therapeutic Program, <http://dtp.nci.nih.gov/>). Therefore, compounds that inhibit cellular proliferation, irrespective of the concentration demonstrating inhibition, have potential clinical usefulness.

[00195] To employ the compounds of the invention in a method of screening for compounds that bind to KSP kinesin, the KSP is bound to a support, and a compound of the invention is added to the assay. Alternatively, the compound of the invention is bound to the support and KSP is added. Classes of compounds among which novel binding agents may be sought include specific antibodies, non-natural binding agents identified in screens of chemical libraries, peptide analogs, etc. Of particular interest are screening assays for candidate agents that have a low toxicity for human cells. A wide variety of assays may be used for this purpose, including labeled in vitro protein-protein binding assays, electrophoretic mobility shift assays, immunoassays for protein binding, functional assays (phosphorylation assays, etc.) and the like.

[00196] The determination of the binding of the compound of the invention to KSP may be done in a number of ways. In one embodiment, the compound is labeled, for example, with a fluorescent or radioactive moiety, and binding is determined directly. For example, this may be done by attaching all or a portion of KSP to a solid support, adding a labeled test compound (for example a compound of the invention in which at least one atom has been replaced by a detectable isotope), washing off excess reagent, and determining whether the amount of the label is that present on the solid support.

[00197] By "labeled" herein is meant that the compound is either directly or indirectly labeled with a label which provides a detectable signal, e.g., radioisotope, fluorescent tag, enzyme, antibodies, particles such as magnetic particles, chemiluminescent tag, or specific binding molecules, etc. Specific binding molecules include pairs, such as biotin and streptavidin, digoxin and antidigoxin etc. For the specific binding members, the complementary member would normally be labeled with a molecule which provides for detection, in accordance with known procedures, as outlined above. The label can directly or

indirectly provide a detectable signal.

[00198] In some embodiments, only one of the components is labeled. For example, the kinesin proteins may be labeled at tyrosine positions using ^{125}I , or with fluorophores. Alternatively, more than one component may be labeled with different labels; using ^{125}I for the proteins, for example, and a fluorophor for the antimetabolic agents.

[00199] The compounds of the invention may also be used as competitors to screen for additional drug candidates. "Candidate agent" or "drug candidate" or grammatical equivalents as used herein describe any molecule, e.g., protein, oligopeptide, small organic molecule, polysaccharide, polynucleotide, etc., to be tested for bioactivity. They may be capable of directly or indirectly altering the cellular proliferation phenotype or the expression of a cellular proliferation sequence, including both nucleic acid sequences and protein sequences. In other cases, alteration of cellular proliferation protein binding and/or activity is screened. Screens of this sort may be performed either in the presence or absence of microtubules. In the case where protein binding or activity is screened, particular embodiments exclude molecules already known to bind to that particular protein, for example, polymer structures such as microtubules, and energy sources such as ATP. Particular embodiments of assays herein include candidate agents which do not bind the cellular proliferation protein in its endogenous native state termed herein as "exogenous" agents. In another embodiment, exogenous agents further exclude antibodies to KSP.

[00200] Candidate agents can encompass numerous chemical classes, though typically they are small organic compounds having a molecular weight of more than 100 and less than about 2,500 daltons. Candidate agents comprise functional groups necessary for structural interaction with proteins, particularly hydrogen bonding and lipophilic binding, and typically include at least an amine, carbonyl-, hydroxyl-, ether, or carboxyl group, generally at least two of the functional chemical groups. The candidate agents often comprise cyclical carbon or heterocyclic structures and/or aromatic or polyaromatic structures substituted with one or more of the above functional groups. Candidate agents are also found among biomolecules including peptides, saccharides, fatty acids, steroids, purines, pyrimidines, derivatives, structural analogs or combinations thereof.

[00201] Candidate agents are obtained from a wide variety of sources including libraries of synthetic or natural compounds. For example, numerous means are available for random and directed synthesis of a wide variety of organic compounds and biomolecules, including expression of randomized oligonucleotides. Alternatively, libraries of natural

compounds in the form of bacterial, fungal, plant and animal extracts are available or readily produced. Additionally, natural or synthetically produced libraries and compounds are readily modified through conventional chemical, physical and biochemical means. Known pharmacological agents may be subjected to directed or random chemical modifications, such as acylation, alkylation, esterification, and/or amidification to produce structural analogs.

[00202] Competitive screening assays may be done by combining KSP and a drug candidate in a first sample. A second sample comprises a compound of the present invention, KSP and a drug candidate. This may be performed in either the presence or absence of microtubules. The binding of the drug candidate is determined for both samples, and a change, or difference in binding between the two samples indicates the presence of a drug candidate capable of binding to KSP and potentially inhibiting its activity. That is, if the binding of the drug candidate is different in the second sample relative to the first sample, the drug candidate is capable of binding to KSP.

[00203] In a particular embodiment, the binding of the candidate agent to KSP is determined through the use of competitive binding assays. In this embodiment, the competitor is a binding moiety known to bind to KSP, such as an antibody, peptide, binding partner, ligand, etc. Under certain circumstances, there may be competitive binding as between the candidate agent and the binding moiety, with the binding moiety displacing the candidate agent.

[00204] In one embodiment, the candidate agent is labeled. Either the candidate agent, or the competitor, or both, is added first to KSP for a time sufficient to allow binding, if present. Incubations may be performed at any temperature which facilitates optimal activity, typically between 4 and 40°C.

[00205] Incubation periods are selected for optimum activity, but may also be optimized to facilitate rapid high throughput screening. Typically between 0.1 and 1 hour will be sufficient. Excess reagent is generally removed or washed away. The second component is then added, and the presence or absence of the labeled component is followed, to indicate binding.

[00206] In another embodiment, the competitor is added first, followed by the candidate agent. Displacement of the competitor is an indication the candidate agent is binding to KSP and thus is capable of binding to, and potentially inhibiting, the activity of KSP. In this embodiment, either component can be labeled. Thus, for example, if the competitor is labeled, the presence of label in the wash solution indicates displacement by the

agent. Alternatively, if the candidate agent is labeled, the presence of the label on the support indicates displacement.

[00207] In an alternative embodiment, the candidate agent is added first, with incubation and washing, followed by the competitor. The absence of binding by the competitor may indicate the candidate agent is bound to KSP with a higher affinity. Thus, if the candidate agent is labeled, the presence of the label on the support, coupled with a lack of competitor binding, may indicate the candidate agent is capable of binding to KSP.

[00208] Inhibition is tested by screening for candidate agents capable of inhibiting the activity of KSP comprising the steps of combining a candidate agent with KSP, as above, and determining an alteration in the biological activity of KSP. Thus, in this embodiment, the candidate agent should both bind to KSP (although this may not be necessary), and alter its biological or biochemical activity as defined herein. The methods include both in vitro screening methods and in vivo screening of cells for alterations in cell cycle distribution, cell viability, or for the presence, morphology, activity, distribution, or amount of mitotic spindles, as are generally outlined above.

[00209] Alternatively, differential screening may be used to identify drug candidates that bind to the native KSP, but cannot bind to modified KSP.

[00210] Positive controls and negative controls may be used in the assays. Suitably all control and test samples are performed in at least triplicate to obtain statistically significant results. Incubation of all samples is for a time sufficient for the binding of the agent to the protein. Following incubation, all samples are washed free of non-specifically bound material and the amount of bound, generally labeled agent determined. For example, where a radiolabel is employed, the samples may be counted in a scintillation counter to determine the amount of bound compound.

[00211] A variety of other reagents may be included in the screening assays. These include reagents like salts, neutral proteins, e.g., albumin, detergents, etc which may be used to facilitate optimal protein-protein binding and/or reduce non-specific or background interactions. Also reagents that otherwise improve the efficiency of the assay, such as protease inhibitors, nuclease inhibitors, anti-microbial agents, etc., may be used. The mixture of components may be added in any order that provides for the requisite binding.

Administration

[00212] Accordingly, the compounds of the invention are administered to cells. By

"administered" herein is meant administration of a therapeutically effective dose of a compound of the invention to a cell either in cell culture or in a patient. By "therapeutically effective dose" herein is meant a dose that produces the effects for which it is administered. The exact dose will depend on the purpose of the treatment, and will be ascertainable by one skilled in the art using known techniques. As is known in the art, adjustments for systemic versus localized delivery, age, body weight, general health, sex, diet, time of administration, drug interaction and the severity of the condition may be necessary, and will be ascertainable with routine experimentation by those skilled in the art. By "cells" herein is meant any cell in which mitosis or meiosis can be altered.

[00213] A "patient" for the purposes of the present invention includes both humans and other animals, particularly mammals, and other organisms. Thus the methods are applicable to both human therapy and veterinary applications. In a particular embodiment the patient is a mammal, and more particularly, the patient is human.

[00214] Compounds of the invention having the desired pharmacological activity may be administered, especially as a pharmaceutically acceptable composition comprising an pharmaceutical excipient, to a patient, as described herein. Depending upon the manner of introduction, the compounds may be formulated in a variety of ways as discussed below. The concentration of therapeutically active compound in the formulation may vary from about 0.1-100 wt.%.

[00215] The agents may be administered alone or in combination with other treatments, i.e., radiation, or other chemotherapeutic agents such as the taxane class of agents that appear to act on microtubule formation or the camptothecin class of topoisomerase I inhibitors. When used, other chemotherapeutic agents may be administered before, concurrently, or after administration of a compound of the present invention. In one aspect of the invention, a compound of the present invention is co-administered with one or more other chemotherapeutic agents. By "co-administer" it is meant that the present compounds are administered to a patient such that the present compounds as well as the co-administered compound may be found in the patient's bloodstream at the same time, regardless when the compounds are actually administered, including simultaneously.

[00216] The administration of the compounds and compositions of the present invention can be done in a variety of ways, including, but not limited to, orally, subcutaneously, intravenously, intranasally, transdermally, intraperitoneally, intramuscularly, intrapulmonary, vaginally, rectally, or intraocularly. In some instances, for example, in the

treatment of wounds and inflammation, the compound or composition may be directly applied as a solution or spray.

[00217] Pharmaceutical dosage forms include a compound of formula I or a pharmaceutically acceptable salt, solvate, or solvate of a salt thereof, and one or more pharmaceutical excipients. As is known in the art, pharmaceutical excipients are secondary ingredients which function to enable or enhance the delivery of a drug or medicine in a variety of dosage forms (e.g.: oral forms such as tablets, capsules, and liquids; topical forms such as dermal, ophthalmic, and otic forms; suppositories; injectables; respiratory forms and the like). Pharmaceutical excipients include inert or inactive ingredients, synergists or chemicals that substantively contribute to the medicinal effects of the active ingredient. For example, pharmaceutical excipients may function to improve flow characteristics, product uniformity, stability, taste, or appearance, to ease handling and administration of dose, for convenience of use, or to control bioavailability. While pharmaceutical excipients are commonly described as being inert or inactive, it is appreciated in the art that there is a relationship between the properties of the pharmaceutical excipients and the dosage forms containing them.

[00218] Pharmaceutical excipients suitable for use as carriers or diluents are well known in the art, and may be used in a variety of formulations. See, e.g., Remington's Pharmaceutical Sciences, 18th Edition, A. R. Gennaro, Editor, Mack Publishing Company (1990); Remington: The Science and Practice of Pharmacy, 20th Edition, A. R. Gennaro, Editor, Lippincott Williams & Wilkins (2000); Handbook of Pharmaceutical Excipients, 3rd Edition, A. H. Kibbe, Editor, American Pharmaceutical Association, and Pharmaceutical Press (2000); and Handbook of Pharmaceutical Additives, compiled by Michael and Irene Ash, Gower (1995), each of which is incorporated herein by reference for all purposes.

[00219] Oral solid dosage forms such as tablets will typically comprise one or more pharmaceutical excipients, which may for example help impart satisfactory processing and compression characteristics, or provide additional desirable physical characteristics to the tablet. Such pharmaceutical excipients may be selected from diluents, binders, glidants, lubricants, disintegrants, colors, flavors, sweetening agents, polymers, waxes or other solubility-retarding materials.

[00220] Compositions for intravenous administration will generally comprise intravenous fluids, i.e., sterile solutions of simple chemicals such as sugars, amino acids or electrolytes, which can be easily carried by the circulatory system and assimilated. Such

fluids are prepared with water for injection USP.

[00221] Dosage forms for parenteral administration will generally comprise fluids, particularly intravenous fluids, i.e., sterile solutions of simple chemicals such as sugars, amino acids or electrolytes, which can be easily carried by the circulatory system and assimilated. Such fluids are typically prepared with water for injection USP. Fluids used commonly for intravenous (IV) use are disclosed in Remington, The Science and Practice of Pharmacy [full citation previously provided], and include:

- alcohol, e.g., 5% alcohol (e.g., in dextrose and water ("D/W") or D/W in normal saline solution ("NSS"), including in 5% dextrose and water ("D5/W"), or D5/W in NSS);
- synthetic amino acid such as Aminosyn, FreAmine, Travasol, e.g., 3.5 or 7; 8.5; 3.5, 5.5 or 8.5 % respectively;
- ammonium chloride e.g., 2.14%;
- dextran 40, in NSS e.g., 10% or in D5/W e.g., 10%;
- dextran 70, in NSS e.g., 6% or in D5/W e.g., 6%;
- dextrose (glucose, D5/W) e.g., 2.5-50%;
- dextrose and sodium chloride e.g., 5-20% dextrose and 0.22-0.9% NaCl;
- lactated Ringer's (Hartmann's) e.g., NaCl 0.6%, KCl 0.03%, CaCl₂ 0.02%;
- lactate 0.3%;
- mannitol e.g., 5%, optionally in combination with dextrose e.g., 10% or NaCl e.g., 15 or 20%;
- multiple electrolyte solutions with varying combinations of electrolytes, dextrose, fructose, invert sugar Ringer's e.g., NaCl 0.86%, KCl 0.03%, CaCl₂ 0.033%;
- sodium bicarbonate e.g., 5%;
- sodium chloride e.g., 0.45, 0.9, 3, or 5%;
- sodium lactate e.g., 1/6 M; and
- sterile water for injection

The pH of such IV fluids may vary, and will typically be from 3.5 to 8 as known in the art.

[00222] The compounds, pharmaceutically acceptable salts and solvates of the invention can be administered alone or in combination with other treatments, i.e., radiation, or other therapeutic agents, such as the taxane class of agents that appear to act on microtubule

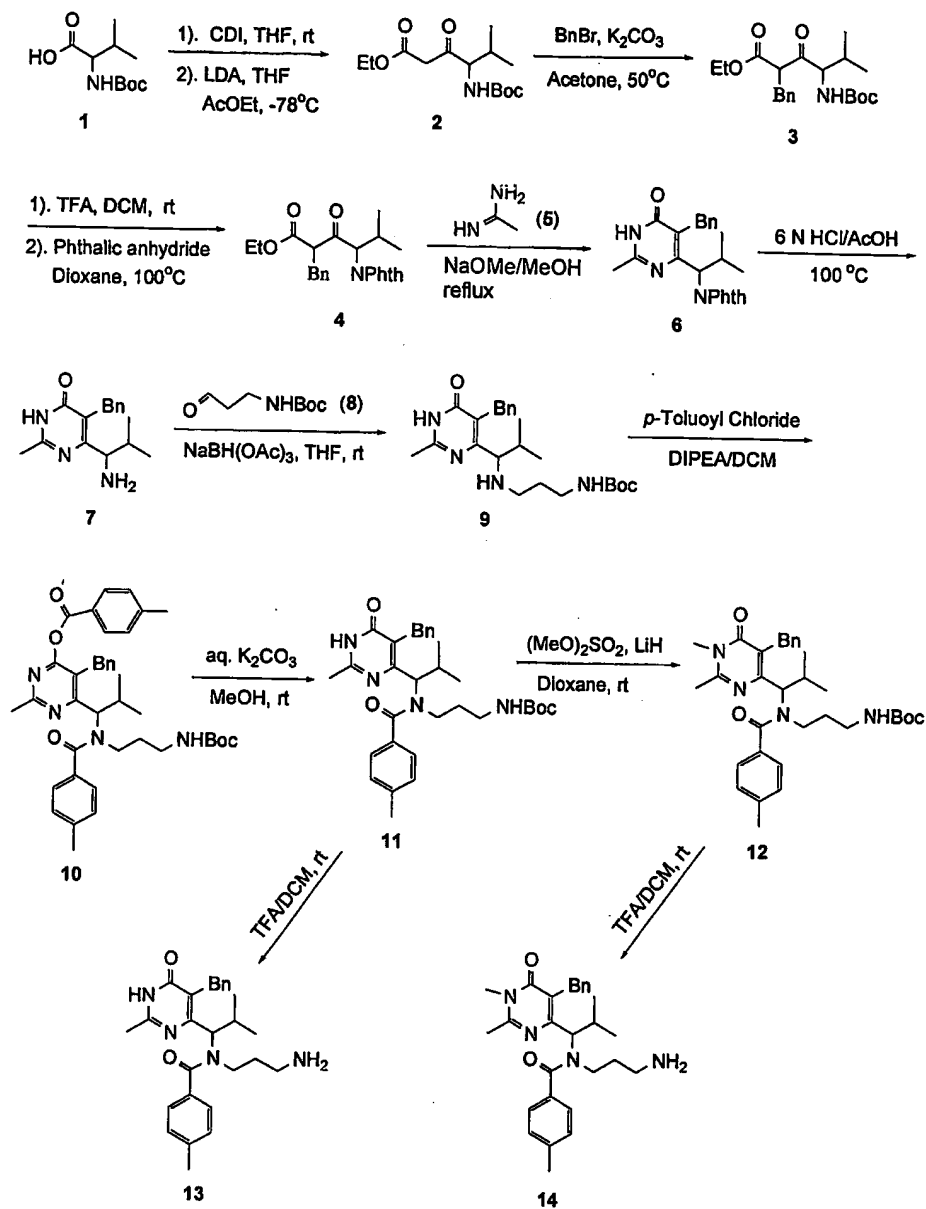
formation or the camptothecin class of topoisomerase I inhibitors. When so-used, other therapeutic agents can be administered before, concurrently (whether in separate dosage forms or in a combined dosage form), or after administration of an active agent of the present invention.

[00223] The following examples serve to more fully describe the manner of using the above-described invention, as well as to set forth the best modes contemplated for carrying out various aspects of the invention. It is understood that these examples in no way serve to limit the true scope of this invention, but rather are presented for illustrative purposes. All publications, including but not limited to patents and patent applications, cited in this specification are herein incorporated by reference as if each individual publication were specifically and individually indicated to be incorporated by reference herein as though fully set forth.

EXAMPLES

[00224] The following examples serve to more fully describe the manner of using the above-described invention, as well as to set forth the best modes contemplated for carrying out various aspects of the invention. It is understood that these examples in no way serve to limit the true scope of this invention, but rather are presented for illustrative purposes. All references cited herein are incorporated by reference in their entirety.

Example 1

**Experimental Section:**

[00225] To a solution of diisoprylamine (18.2 mL, 130 mmol) in anhydrous

tetrahydrofuran (200 mL) at 0 °C was added n-butyllithium (52.0 mL, 2.5 M in THF, 130 mmol) and the resulting mixture was stirred at the same temperature for another hour. After this mixture was cooled to -78 °C, ethyl acetate (12.7 mL, 120 mmol) was added, and the resulting solution was stirred at the same temperature for 1.5 hours.

[00226] Meanwhile, to a solution of N-Boc-Val-OH (8.8 g, 40.0 mmol) in anhydrous tetrahydrofuran (150 mL) at room temperature was added 1,1'-carbonyldiimidazole (6.6g, 40.0 mmol). The resulting mixture was stirred at room temperature for 1.5 hours. This mixture was added to the above freshly prepared ethyl lithium acetate solution at -78 °C. After being stirred for 2 hours at the same temperature, the reaction was quenched with aqueous sodium bicarbonate and extracted with ethyl acetate (3 x 200 mL). The combined organic layers were dried over sodium sulfate and evaporated; the residue (10.8 g) was purified by flash chromatography (silica gel, hexane/ ethyl acetate). The desired product 2 (8.2 g) was isolated, which was characterized using ¹H-NMR and LC/MS (LRMS (MH) *m/z*: 287.35).

[00227] To a mixture of the starting amide 2 (2.87 g, 10.0 mmol) and potassium carbonate (4.14 g) in acetone (150 mL) at room temperature was added benzyl bromide (1.32 mL, 11.0 mmol) The resulting solution was heated to 60 °C for 4 hours until TLC indicated that almost no starting material presented. It was cooled down, diluted with water, and extracted with ethyl acetate. The combined organic layers were dried over sodium sulfate and evaporated; the residue was purified by flash chromatography (silica gel, hexane/ethyl acetate), and the desired product 3 (2.20 g) was isolated as a mixture of diastereoisomers, which was characterized using ¹H-NMR and LC/MS (LRMS (MH) *m/z*: 377.47).

[00228] To a solution of compound 3 (7.0 g, 18.5 mmol) in dichloromethane (100 mL) at 0 °C was added trifluoroacetic acid (40 mL). The resulting solution was stirred at room temperature for 2 hours and then concentrated under reduced pressure. The residue (7.5 g) was dried under high vacuum and used in the next step without further purification.

[00229] To a solution of the above crude residue (7.5 g) in dioxane (200 mL) was added phthalic anhydride (8.1 g, 54.0 mmol), and the resulting solution was heated to 100 °C until LC/MS indicated almost no starting material present. The reaction was cooled down and diluted with saturated aqueous sodium bicarbonate solution. The resulting solution was saturated by sodium chloride and extracted with tetrahydrofuran (3 x 150 mL). The combined organic layers were dried over sodium sulfate and evaporated; the residue was purified by flash chromatography (silica gel, dichloromethane and methanol), and the desired product 4

(4.2 g) was isolated as a mixture of diastereoisomers, which was characterized using ¹H-NMR and LC/MS (LRMS (MH) *m/z*: 407.46).

[00230] To a solution of ester 4 (700 mg, 1.65 mmol) in methanol (10 mL) were added acetamidine 5 (800 mg, 8.5 mmol) and sodium methoxide (10.0 mL, 5.00 mmol, 0.5 M in methanol) successively at room temperature. The resulting solution was heated to 60 °C for 48 hours. It was cooled down and the solvent was removed under reduced pressure, the residue was dissolved in distilled water (20 mL) and saturated with sodium chloride. It was extracted with tetrahydrofuran (3 x 100 mL), and the combined organic layers were dried over sodium sulfate. After evaporation of the solvents, the residue was purified using column chromatography (silica gel, dichloromethane/methanol). The desired pyrimidinone 6 (150 mg) was isolated and characterized using ¹H-NMR and LC/MS (LRMS (MH) *m/z*: 401.46).

[00231] To a solution of pyrimidinone 6 (200 mg, 0.50 mmol) in acetic acid (4.0 mL) at room temperature was added aqueous hydrochloric acid (6 N, 8.0 mL). The resulting solution was stirred at 110 °C for 1 hour and monitored by LC/MS. 24 hours later, the starting material almost disappeared and the resulting solution was cooled down. It was diluted with tetrahydrofuran, neutralized to PH=8 with sodium carbonate, and saturated with sodium chloride. The aqueous phase was extracted with tetrahydrofuran (3 x 80 mL) until no desired product presented in the aqueous phase. The combined organic layers were dried over sodium sulfate and concentrated; the residue was purified on flash chromatography (silica gel, dichloromethane/ methanol) to provide the desired product 7 (70 mg), which was characterized using LC/MS (LRMS (MH) *m/z* 271.36).

[00232] To a solution of pyrimidinone 7 (58 mg, 0.22 mmol) in tetrahydrofuran (5 mL) at room temperature were added sodium triacetoxyborohydride (58 mg, 0.26 mmol) and aldehyde 8 (52 mg, 0.30 mmol) successively. The resulting mixture was stirred at room temperature under nitrogen for 4 hours until almost no starting material presented. It was quenched with water (4 mL), saturated with sodium chloride, and extracted with tetrahydrofuran (3 x 60 mL). The combined organic layers were dried over sodium sulfate and concentrated; the residue was purified by flash chromatography (silica gel, dichloromethane/ methanol) to provide the desired product 9 (65 mg), which was characterized using LC/MS (LRMS (MH) *m/z* 428.57).

[00233] To a solution of pyrimidinone 9 (277 mg, 0.65 mmol) in dichloromethane (20 mL) at 0 °C were added diisopropylethylamine (1.0 mL) and *p*-toluoyl chloride (300 mg, 1.92 mmol), successively. The resulting solution was stirred at room temperature under nitrogen

overnight. The solvents were evaporated and the residue dissolved in methanol (20 mL). Aqueous potassium carbonate solution was added until pH = 8 and the resulting solution were stirred at room temperature for one hour. Most of the solvent was evaporated and the resulting aqueous phase was extracted with ethyl acetate (4 x 50 mL). The combined organic layers were dried over sodium sulfate. After evaporation of the solvents, the residue was purified using flash column chromatography (silica gel, dichloromethane/methanol) to provide the product 11 (300 mg), which was characterized using ¹H-NMR and LC/MS (LRMS (MH) *m/z* 546.70).

[00234] To a solution of pyrimidinone 11 (112 mg, 0.21 mmol) in dioxane (15 mL) were added lithium hydride (6.0 mg, 0.63 mmol) and dimethyl sulfate (40.0 mg, 0.32 mmol), successively. The resulting mixture was stirred at room temperature for 4 hours and at 40 °C for 15 minutes. LC/MS indicated that the starting material disappeared and just one regioisomer produced. The mixture was cooled down and diluted with ethyl acetate. It was then quenched with saturated aqueous sodium bicarbonate solution and extracted with ethyl acetate (3 x 50 mL). The combined organic layers were dried over sodium sulfate. After evaporation of the solvents, the residue 12 (120 mg), which was characterized using LC/MS (LRMS (MH) *m/z* 560.73), was used in the next step without further purification.

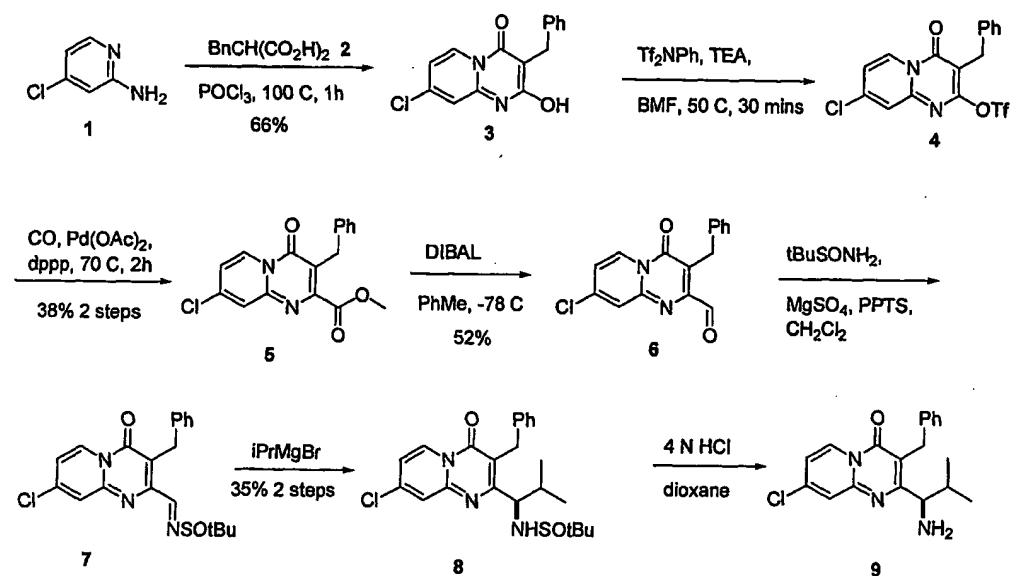
[00235] To a solution of pyrimidinone 11 (54 mg, 0.10 mmol) in dichloromethane (6 mL) at 0 °C was added trifluoroacetic acid (2 mL). The resulting solution was stirred at room temperature for 2 hours and then concentrated under reduced pressure. The residue was dried under high vacuum for 1 hour and dissolved in ethyl acetate (25 mL). It was neutralized with saturated aqueous sodium bicarbonate solution and the aqueous phase was extracted with ethyl acetate (3 x 25 mL). The combined organic layers were dried over sodium sulfate and concentrated; the residue was purified by flash column chromatography (silica gel, methanol/dichloromethane) to provide the desired product 13 (40 mg), which was fully characterized with ¹H-NMR and LC/MS analysis (LRMS (MH) *m/z*: 446.58).

[00236] To a solution of crude pyrimidinone 12 (120 mg) in dichloromethane (6 mL) at 0 °C was added trifluoroacetic acid (2 mL). The resulting solution was stirred at room temperature for 2 hours and then concentrated under reduced pressure. The residue was dried under high vacuum for 1 hour and dissolved in ethyl acetate (25 mL). It was neutralized with saturated aqueous sodium bicarbonate solution and the aqueous phase was extracted with ethyl acetate (3 x 25 mL). The combined organic layers were dried over sodium sulfate and concentrated; the residue was purified by flash column chromatography (silica gel,

methanol/dichloromethane) to provide the desired product **14** (60 mg), which was fully characterized with ^1H -NMR and LC/MS analysis (LRMS (MH) m/z : 460.61).

Example 2

Synthesis of Compounds



[00237] A solution of 2-amino-4-chloropyridine¹ (**1**, 2.22 g, 17.3 mmol), benzyl malonic acid (**2**, 4.04 g, 20.8 mmol), and POCl_3 (35 mL) was maintained at 100 C for 2 hours. The solution was then concentrated and to resulting residue was added EtOAc (10 mL) and ice water (50 mL). The resulting mixture was stirred vigorously for 30 mins at 0 C. The mixture was filtered and the resulting solid rinsed with Et₂O. The resulting solid (3.3 g, 66% yield) was >95% pure by ^1H NMR and LCMS analysis used without further purification. LRMS (MH) m/z 287.1.

[00238] A solution of pyridopyrimidinone **3** (1.04 g, 3.66 mmol), N-phenylbis(trifluoromethane-sulfonamide) (2.71 g, 7.23 mmol), triethylamine (2.0 mL, 14.5 mmol), and DMF (18 mL) was maintained at 50 C for 30 mins. The solution was then diluted with EtOAc (150 mL) and washed with brine (3 x 50 mL). The organic layer was dried (MgSO_4), filtered, and concentrated. The resulting residue was used without further

purification. LRMS (MH) m/z 418.9.

[00239] A Parr bottle containing a solution of crude pyridopyrimidinone triflate 4 (~3.66 mmol), Pd(OAc)₂ (81 mg, 0.36 mmol), 1,3-bis(diphenylphosphino)propane (150 mg, 0.36), triethylamine (2.0 mL, 14.5 mmol), and MeOH (18 mL) was purged with carbon monoxide for 10 minutes. The apparatus was then sealed, pressurized with carbon monoxide (50 psi), and heated to 70 C for 2 hours. The solution was then concentrated and the crude residue was diluted with hexanes:EtOAc (3:1, 100 mL) and the resulting solution was passed through a plug of silica gel. The plug was rinsed with additional hexanes:EtOAc (3:1, 100 mL). The resulting residue was purified by flash column chromatography (3:1 hexanes:EtOAc) to yield 450 mg (38%, 2 steps) of 5. LRMS (MH) m/z 329.0.

[00240] Diisobutylaluminum hydride (1.5 M in PhMe, 1.3 mL) was added dropwise to a -78 C solution of pyridopyrimidinone ester 5 (241 mg, 0.67 mmol) and toluene (11 mL). After 15 mins, the reaction solution was quenched with 1 N HCl (20 mL) and EtOAc (40 mL). The layers were separated and the organic layer was washed with brine (10 mL). The organic layer was dried (MgSO₄), filtered, and concentrated. The resulting residue was purified by flash column chromatography (3:1 hexanes:EtOAc) to yield 105 mg (52%) of 6. LRMS (MH) m/z 299.1.

[00241] A mixture of pyridopyrimidinone aldehyde 6 (105 mg, 0.35 mmol), (S)-(-)-2-methyl-2-propanesulfinamide (68 mg, 0.56 mmol), pyridinium *p*-toluenesulfonate (10 mg, 0.04 mmol), anhydrous magnesium sulfate (210 mg, 1.75 mmol), and CH₂Cl₂ (1 mL) was maintained at r.t. for 18 hours. The reaction mixture was diluted with EtOAc (15 mL) and passed through a plug of silica gel. The plug was rinsed with an additional portion of EtOAc (15 mL). The filtrate was concentrated, and the crude residue was dissolved in 1:1 hexanes:EtOAc (15 mL) and filtered through a plug of silica gel. The plug was rinsed with an additional portion of 1:1 hexanes:EtOAc (15 mL) and the filtrate was concentrate to provide 7 as yellow solid that was used without additional purification. LRMS (MH) m/z 402.1.

[00242] Isopropyl magnesium bromide (1.0 M in THF, 0.7 mL) was added dropwise to a -78 C solution of crude pyridopyrimidinone 7 (~0.35 mmol) and THF (3.5 mL). After 30 mins, the reaction solution was quenched with 1 N HCl (5 mL) and EtOAc (25 mL). The layers were separated and the organic layer was washed with brine (10 mL). The organic layer was dried (MgSO₄), filtered, and concentrated. The resulting residue was purified by flash

¹ Gudmundsson, K. S., Hinkley, J. M., Brieger, M. S., Drach, J. C., Townsend, L. B. *Synthetic Communications*,

column chromatography (1:1 hexanes:EtOAc; 20:1 CH₂Cl₂:MeOH) to yield 55 mg (35%, 2 steps) of 8. LRMS (MH) *m/z* 446.1.

[00243] A solution of pyridopyrimidine 8 (55 mg, 0.12 mmol), MeOH (2 mL), and 4 N HCl in dioxane (2 mL) was maintained at r.t. for 30 mins. The reaction solution was then concentrated. The resulting residue was dissolved in EtOAc (20 mL) and washed with 1 N NaOH (5 mL) and brine (5 mL). The layers were separated and the organic layer was dried (MgSO₄), filtered, and concentrated. The resulting residue was used without further purification. LRMS (MH) *m/z* 342.1.

Example 3

Inhibition of Cellular Viability in Tumor Cell Lines Treated with KSP Inhibitors.

[00244] Materials and Solutions:

- Cells: SKOV3, Ovarian Cancer (human).
- Media: Phenol Red Free RPMI + 5% Fetal Bovine Serum + 2mM L-glutamine.
- Colorimetric Agent for Determining Cell Viability: Promega MTS tetrazolium compound.
- Control Compound for max cell kill: Topotecan, 1 μ M.

Procedure: Day 1 - Cell Plating:

[00245] Adherent SKOV3 cells are washed with 10mLs of PBS followed by the addition of 2mLs of 0.25% trypsin and incubation for 5 minutes at 37°C. The cells are rinsed from the flask using 8 mL of media (phenol red-free RPMI+ 5%FBS) and transferred to fresh flask. Cell concentration is determined using a Coulter counter and the appropriate volume of cells to achieve 1000 cells/100 μ L is calculated. 100 μ L of media cell suspension (adjusted to 1000 cells/100 μ L) is added to all wells of 96-well plates, followed by incubation for 18 to 24 hours at 37°C, 100% humidity, and 5% CO₂, allowing the cells to adhere to the plates.

Procedure: Day 2 – Compound Addition:

[00246] To one column of the wells of an autoclaved assay block are added an initial

2.5 μL of test compound(s) at 400X the highest desired concentration. 1.25 μL of 400X (400 μM) Topotecan is added to other wells (ODs from these wells are used to subtract out for background absorbance of dead cells and vehicle). 500 μL of media without DMSO are added to the wells containing test compound, and 250 μL to the Topotecan wells. 250 μL of media + 0.5% DMSO is added to all remaining wells, into which the test compound(s) are serially diluted. By row, compound-containing media is replica plated (in duplicate) from the assay block to the corresponding cell plates. The cell plates are incubated for 72 hours at 37°C, 100% humidity, and 5% CO₂.

Procedure: Day 4 – MTS Addition and OD Reading:

[00247] The plates are removed from the incubator and 40 μL MTS / PMS is added to each well. Plates are then incubated for 120 minutes at 37°C, 100% humidity, 5% CO₂, followed by reading the ODs at 490nm after a 5 second shaking cycle in a ninety-six well spectrophotometer.

Data Analysis

[00248] The normalized % of control (absorbance- background) is calculated and an XLfit is used to generate a dose-response curve from which the concentration of compound required to inhibit viability by 50% is determined. The compounds of the present invention show activity when tested by this method.

Example 4

Monopolar Spindle Formation following Application of a KSP Inhibitor

[00249] Human tumor cells Skov-3 (ovarian) were plated in 96-well plates at densities of 4,000 cells per well, allowed to adhere for 24 hours, and treated with various concentrations of the test compounds for 24 hours. Cells were fixed in 4% formaldehyde and stained with antitubulin antibodies (subsequently recognized using fluorescently-labeled secondary antibody) and Hoechst dye (which stains DNA).

[00250] Visual inspection revealed that the compounds caused cell cycle arrest in the prometaphase stage of mitosis. DNA was condensed and spindle formation had initiated, but arrested cells uniformly displayed monopolar spindles, indicating that there was an inhibition of spindle pole body separation. Microinjection of anti-KSP antibodies also causes mitotic

arrest with arrested cells displaying monopolar spindles.

Example 5

Inhibition of Cellular Proliferation in Tumor Cell Lines Treated with KSP Inhibitors.

[00251] Cells were plated in 96-well plates at densities from 1000-2500 cells/well of a 96-well plate and allowed to adhere/grow for 24 hours. They were then treated with various concentrations of drug for 48 hours. The time at which compounds are added is considered T_0 . A tetrazolium-based assay using the reagent 3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium (MTS) (I.S. Patent No. 5,185,450) (see Promega product catalog #G3580, CellTiter 96® AQueous One Solution Cell Proliferation Assay) was used to determine the number of viable cells at T_0 and the number of cells remaining after 48 hours compound exposure. The number of cells remaining after 48 hours was compared to the number of viable cells at the time of drug addition, allowing for calculation of growth inhibition.

[00252] The growth over 48 hours of cells in control wells that had been treated with vehicle only (0.25% DMSO) is considered 100% growth and the growth of cells in wells with compounds is compared to this. KSP inhibitors inhibited cell proliferation in human ovarian tumor cell lines (SKOV-3).

[00253] A GI_{50} was calculated by plotting the concentration of compound in μM vs the percentage of cell growth of cell growth in treated wells. The GI_{50} calculated for the compounds is the estimated concentration at which growth is inhibited by 50% compared to control, i.e., the concentration at which:

$$100 \times [(Treated_{48} - T_0) / (Control_{48} - T_0)] = 50.$$

[00254] All concentrations of compounds are tested in duplicate and controls are averaged over 12 wells. A very similar 96-well plate layout and GI_{50} calculation scheme is used by the National Cancer Institute (see Monks, et al., J. Natl. Cancer Inst. 83:757-766 (1991)). However, the method by which the National Cancer Institute quantitates cell number does not use MTS, but instead employs alternative methods.

Example 6

Calculation of IC_{50} :

[00255] Measurement of a composition's IC_{50} for KSP activity uses an ATPase assay. The following solutions are used: Solution 1 consists of 3 mM phosphoenolpyruvate

potassium salt (Sigma P-7127), 2 mM ATP (Sigma A-3377), 1 mM IDTT (Sigma D-9779), 5 μ M paclitaxel (Sigma T-7402), 10 ppm antifoam 289 (Sigma A-8436), 25 mM Pipes/KOH pH 6.8 (Sigma P6757), 2 mM MgCl₂ (VWR JT400301), and 1 mM EGTA (Sigma E3889). Solution 2 consists of 1 mM NADH (Sigma N8129), 0.2 mg/ml BSA (Sigma A7906), pyruvate kinase 7U/ml, L-lactate dehydrogenase 10 U/ml (Sigma P0294), 100 nM KSP motor domain, 50 μ g/ml microtubules, 1 mM DTT (Sigma D9779), 5 μ M paclitaxel (Sigma T-7402), 10 ppm antifoam 289 (Sigma A-8436), 25 mM Pipes/KOH pH 6.8 (Sigma P6757), 2 mM MgCl₂ (VWR JT4003-01), and 1 mM EGTA (Sigma E3889). Serial dilutions (8-12 two-fold dilutions) of the composition are made in a 96-well microtiter plate (Corning Costar 3695) using Solution 1. Following serial dilution each well has 50 μ l of Solution 1. The reaction is started by adding 50 μ l of solution 2 to each well. This may be done with a multichannel pipettor either manually or with automated liquid handling devices. The microtiter plate is then transferred to a microplate absorbance reader and multiple absorbance readings at 340 nm are taken for each well in a kinetic mode. The observed rate of change, which is proportional to the ATPase rate, is then plotted as a function of the compound concentration. For a standard IC₅₀ determination the data acquired is fit by the following four parameter equation using a nonlinear fitting program (e.g., Grafit 4):

$$y = \frac{\text{Range}}{1 + \left(\frac{x}{\text{IC}_{50}} \right)^s} + \text{Background}$$

where y is the observed rate and x the compound concentration.

[00256] Other compounds of this class were found to inhibit cell proliferation, although GI₅₀ values varied. GI₅₀ values for the compounds tested ranged from 200 nM to greater than the highest concentration tested. By this we mean that although most of the compounds that inhibited KSP activity biochemically did inhibit cell proliferation, for some, at the highest concentration tested (generally about 20 μ M), cell growth was inhibited less than 50%. Many of the compounds have GI₅₀ values less than 10 μ M, and several have GI₅₀ values less than 1 μ M. Anti-proliferative compounds that have been successfully applied in the clinic to treatment of cancer (cancer chemotherapeutics) have GI₅₀'s that vary greatly. For example, in A549 cells, paclitaxel GI₅₀ is 4 nM, doxorubicin is 63 nM, 5-fluorouracil is 1 μ M, and hydroxyurea is 500 μ M (data provided by National Cancer Institute, Developmental Therapeutic Program, <http://dtp.nci.nih.gov/>). Therefore, compounds that inhibit cellular

proliferation at virtually any concentration may be useful. However, preferably, compounds will have GI_{50} values of less than 1 mM. More preferably, compounds will have GI_{50} values of less than 20 μ M. Even more preferably, compounds will have GI_{50} values of less than 10 μ M. Further reduction in GI_{50} values may also be desirable, including compounds with GI_{50} values of less than 1 μ M. Some of the compounds of the invention inhibit cell proliferation with GI_{50} values from below 200 nM to below 10 nM.